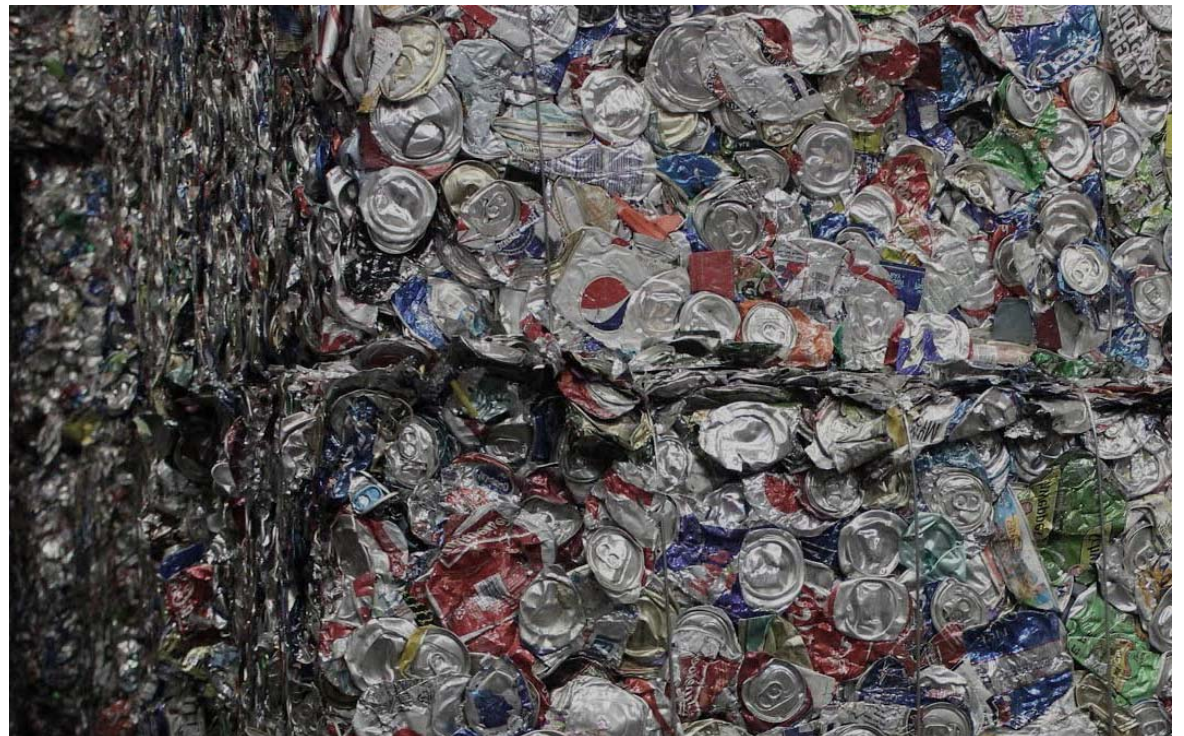




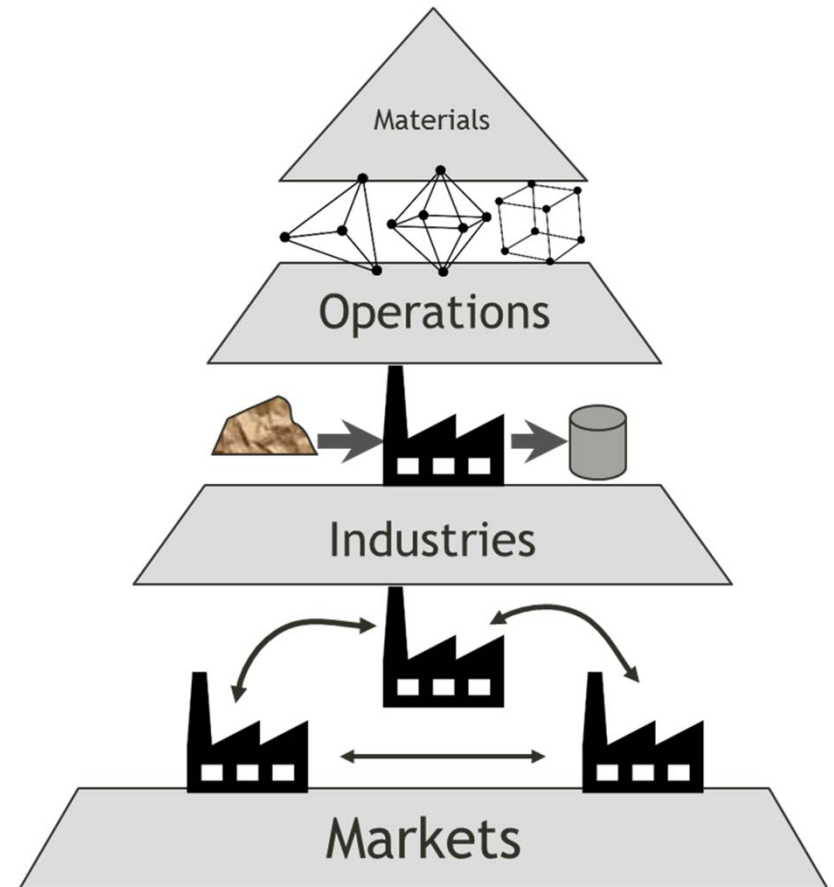
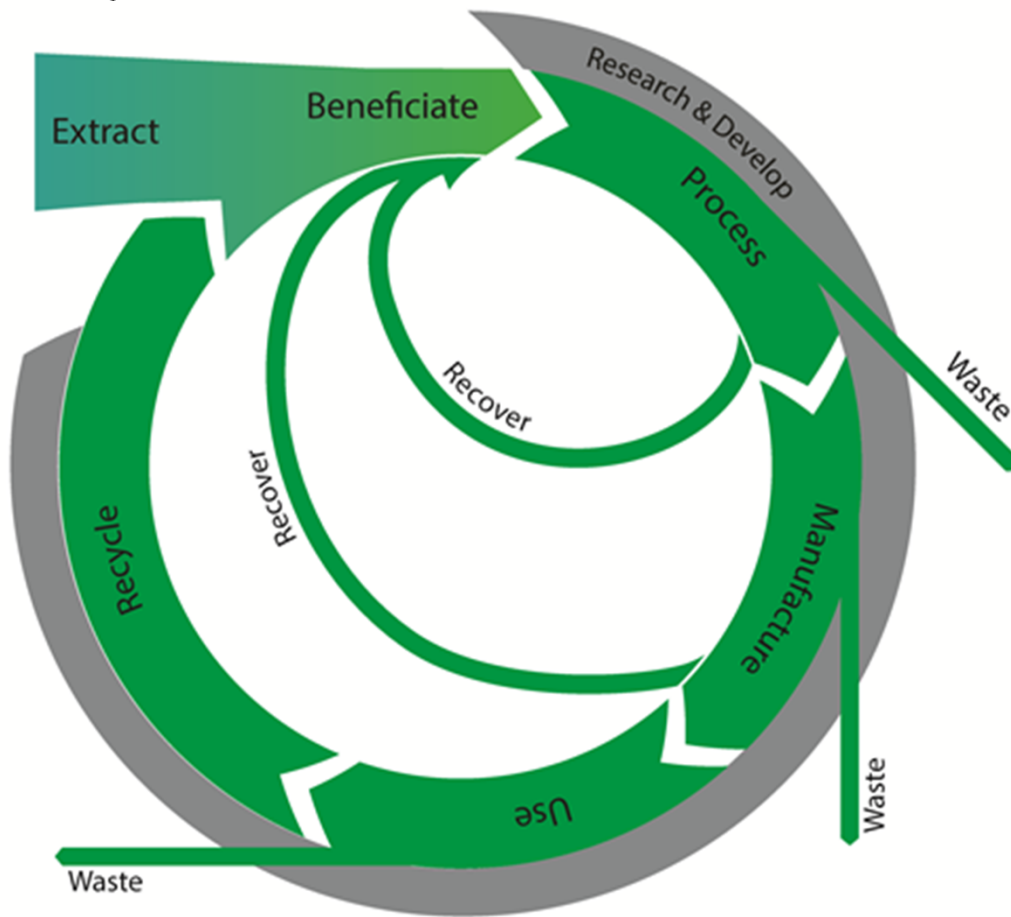
Elsa Olivetti
Department of
Materials Science &
Engineering

Informing design of
resource-effective
materials, processes
and systems



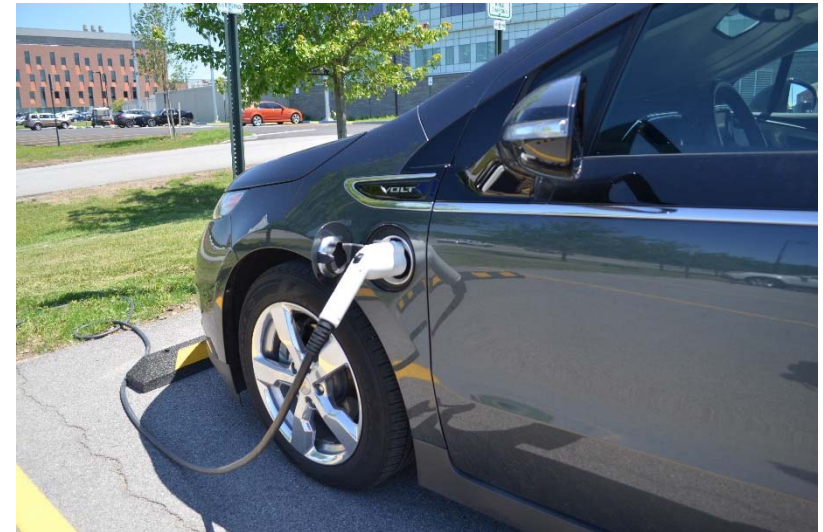
Resource-effective decision making for design of materials, operations, industries, and systems

Materials do not exist in isolation, they are part of complex networks.



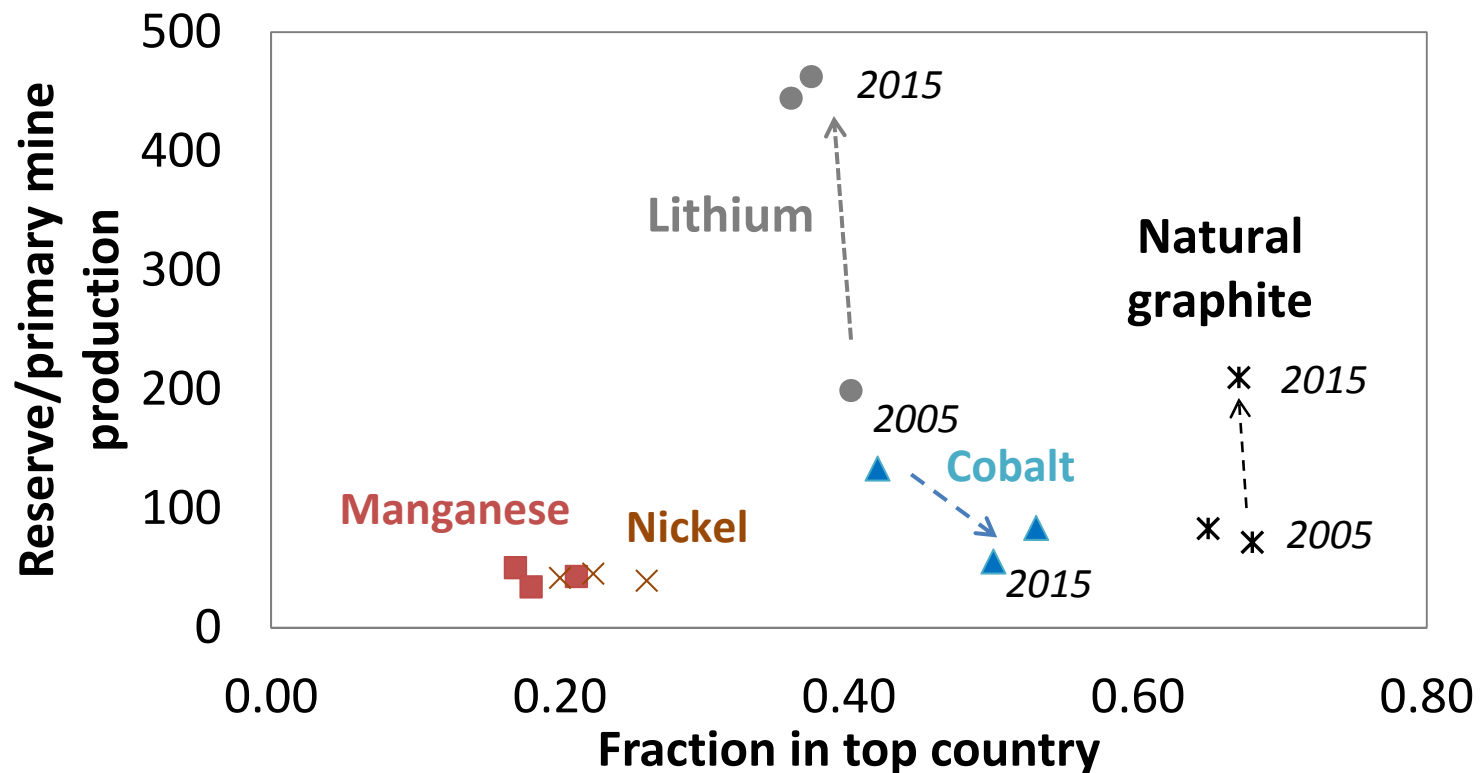
Analysis of Potential Supply Chain Bottlenecks in Metals for Li-ion batteries

Cobalt 27 Co 58.933	Lithium 3 Li 6.941	Carbon 6 C 12.011
Mangan 25 Mn 54.938		Nickel 28 Ni 58.693



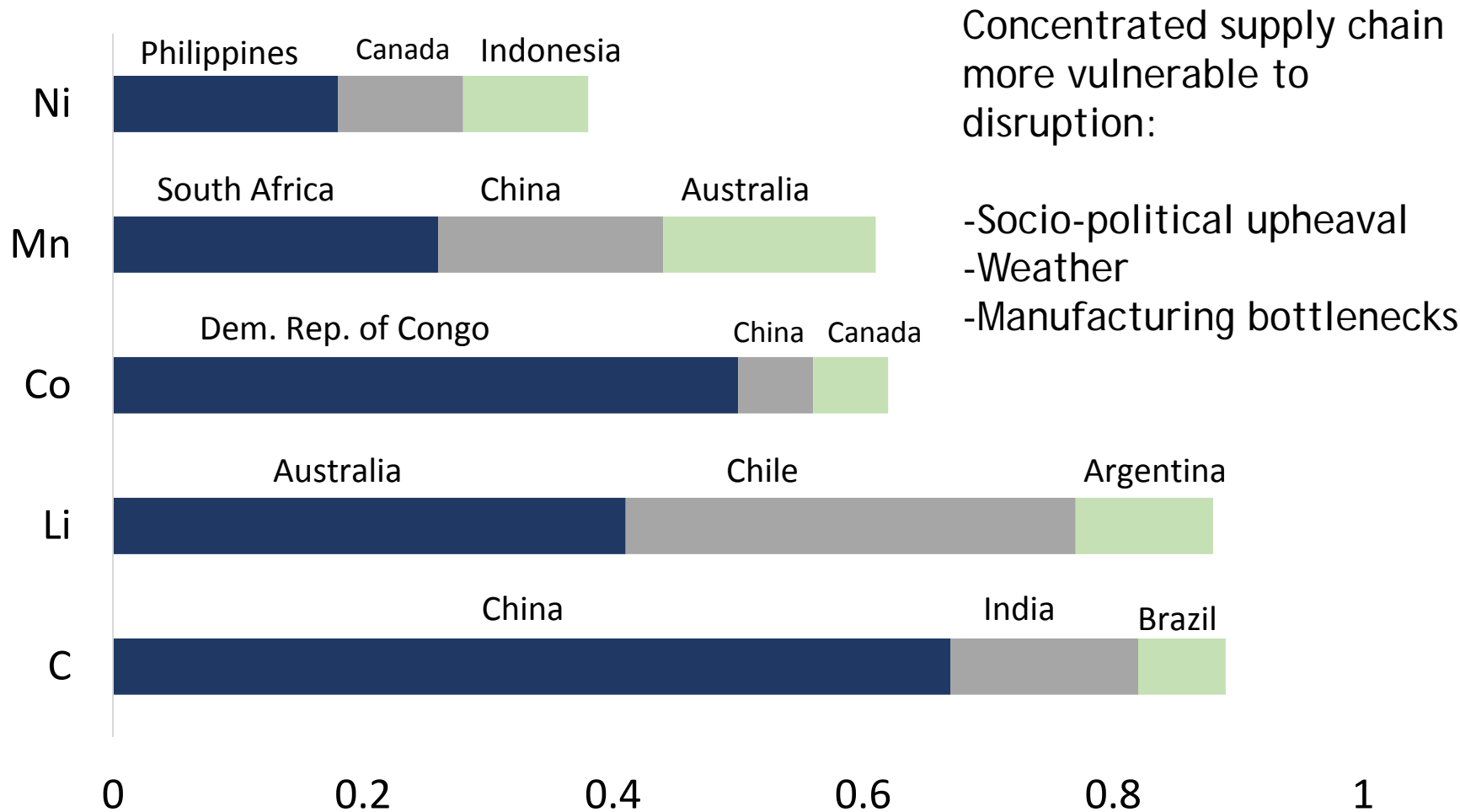
Joule 2017

Resource restrictions for relevant materials



Static depletion above 30 years for all
Ni and Mn index is relatively constant indicating that the economics of demand drive the supply towards continued economical extraction.

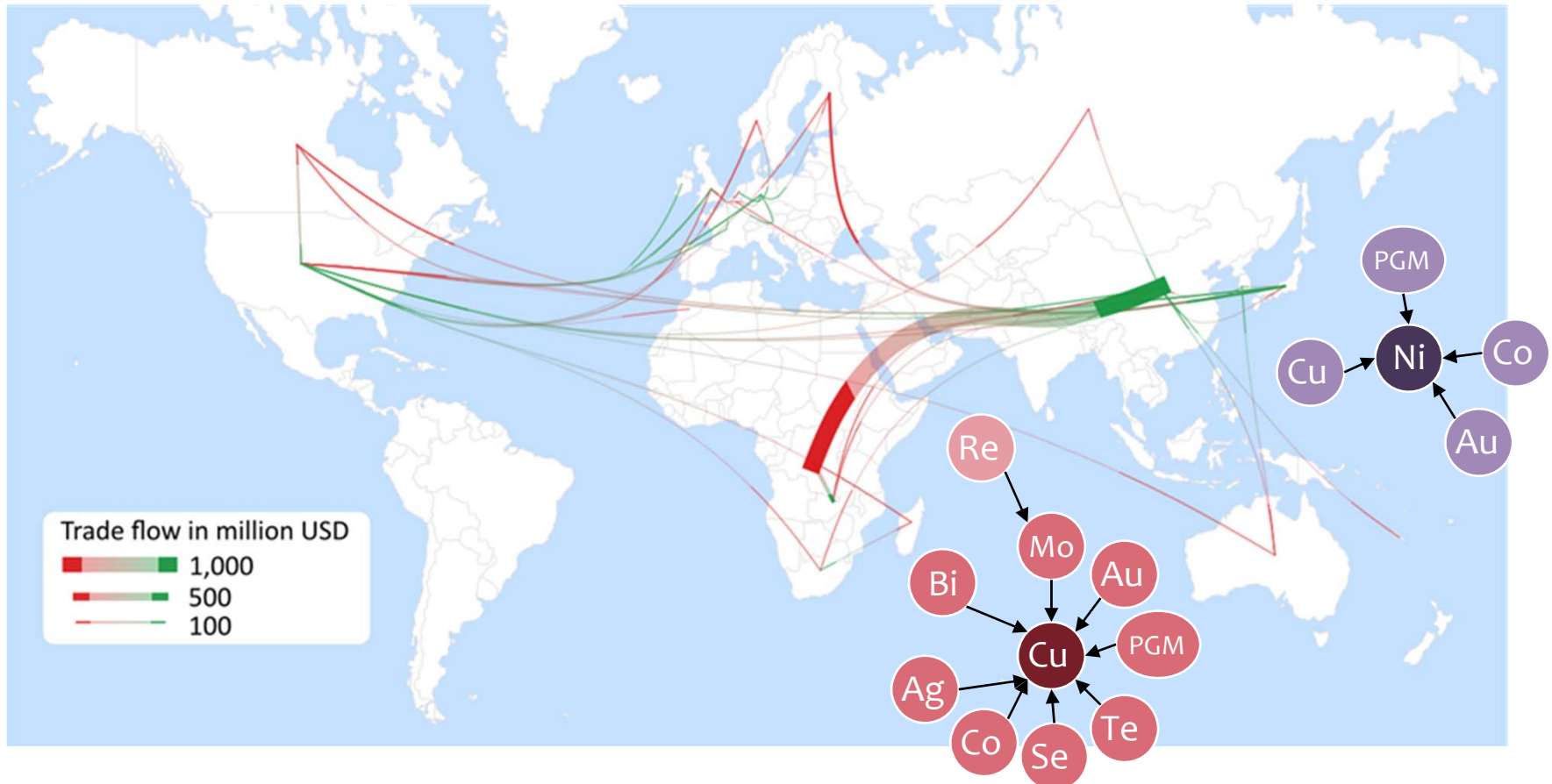
Geographic concentration for relevant materials



Concentrated supply chain more vulnerable to disruption:

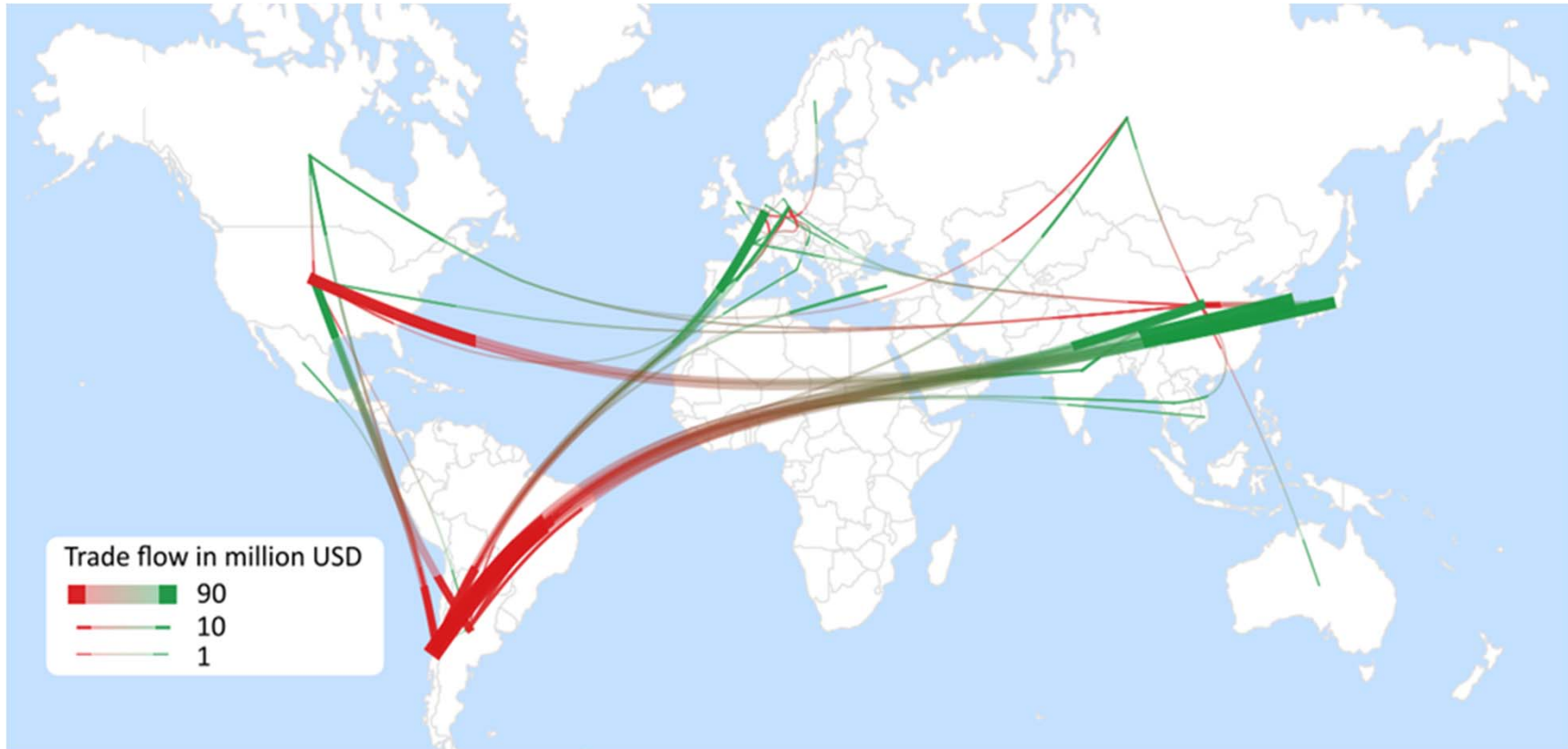
- Socio-political upheaval
- Weather
- Manufacturing bottlenecks

Cobalt supply chain focuses on few dominant players



Co globally concentrated in mining, but it is also geographically concentrated in refining

Lithium supply chain is more diversified



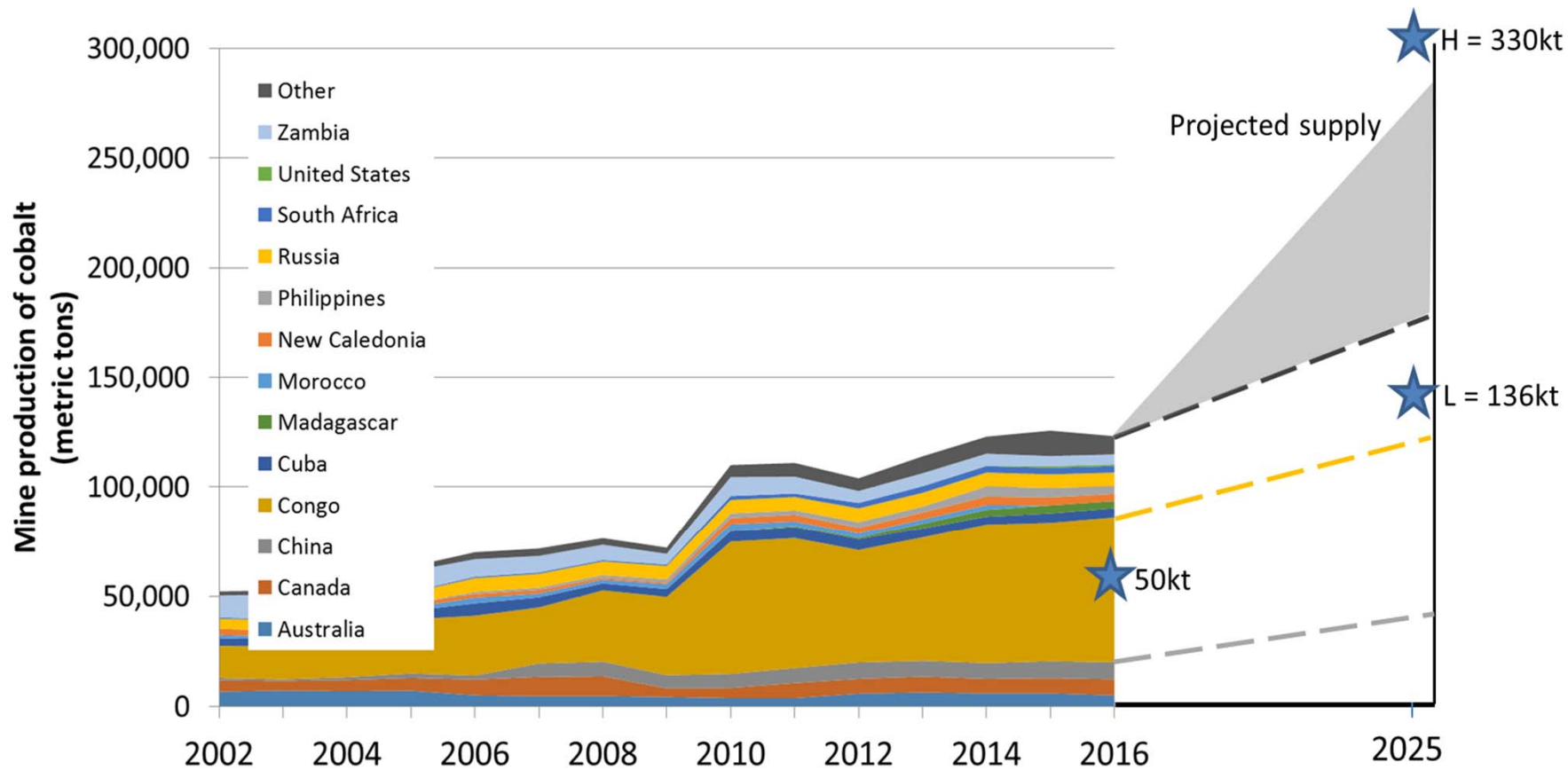
Lithium carbonate recovered via multiple routes and geographically less concentrated
Resource and reserve estimates are still expanding

Amount of material needed per kWh varies by chemistry

	Use	Li	Co	Ni	Mn	Graphitic carbon
Lithium cobalt oxide	electronics	0.113	0.959	0	0	~1.2*
Lithium nickel cobalt aluminum oxide	Auto, grid, other	0.112	0.143	0.759	0	
Lithium nickel manganese cobalt oxide NMC-111		0.139	0.394	0.392	0.367	
NMC-622		0.126	0.214	0.641	0.200	
NMC-811		0.111	0.094	0.750	0.088	

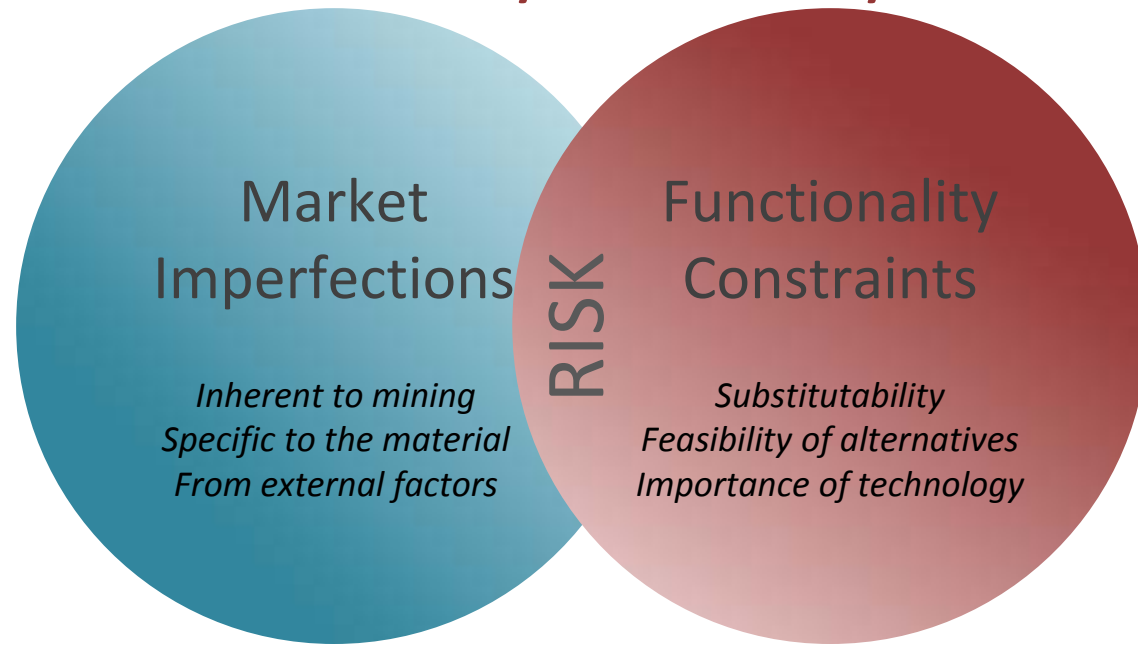
Current metal required in kg/kWh
 * literature values

Focus on cobalt demand and supply



What is materials criticality?

Criticality \neq Scarcity!



Criticality

Risk deemed too high by a decision-maker

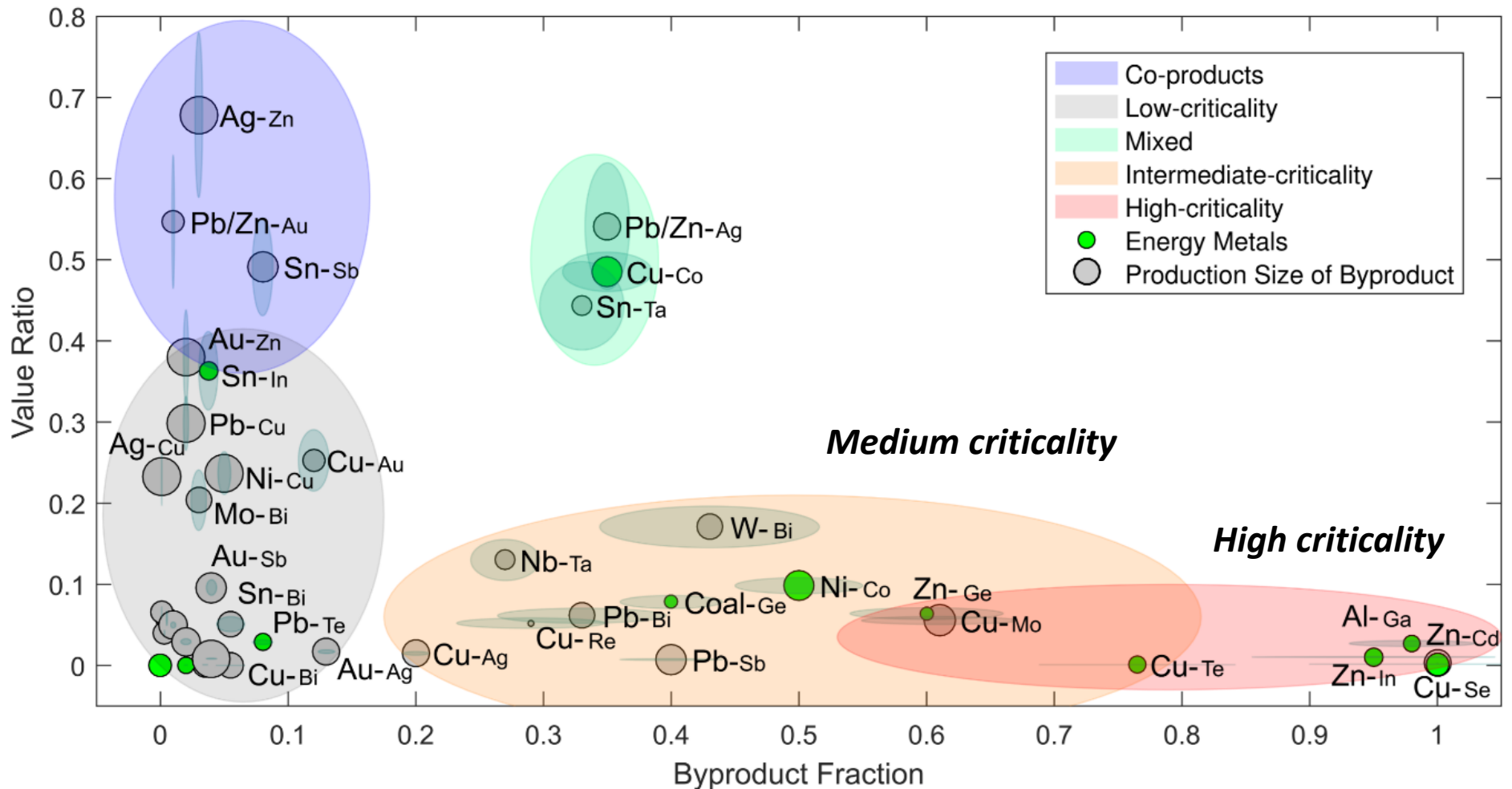
Materials availability:

Byproduct dependency used as metric of criticality

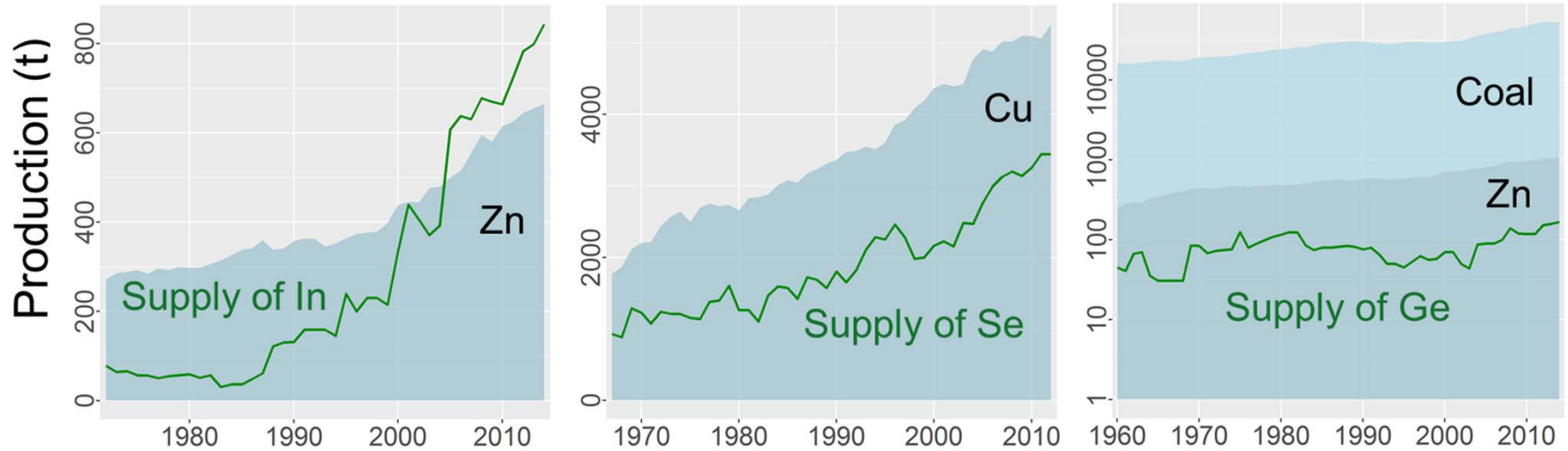
Indicator	Relation with supply risk ^a	Frequency of use	Means of measurement/units
Country concentration of production	Direct	12	Herfindahl–Hirschman-Index
Country governance	Dep. on def.	10	Qualitative, index
Depletion time	Inverse	9	Years
By-product dependency	Direct	7	%
Company concentration in mining corporations ^b	Direct	5	Herfindahl–Hirschman-Index
Demand growth ^b	Direct	5	Qualitative, ratio
Import dependence ^b	Direct	3	%, net value
Recycling/recycling potential ^b	Inverse	3	Tons
Substitutability ^b	Inverse	3	Qualitative
Volatility of commodity prices	Direct	2	USD/kg, EUR/kg
Exploration degree	Inverse	1	USD, EUR
Production costs in extraction	Direct	1	USD, EUR
Stock keeping	Inverse	1	%
Market balance	Direct	1	Tons
Mine/refinery capacity	Inverse	1	%
Future market capacity	Inverse	1	%
Investment in mining	Inverse	1	USD/t, EUR/t
Climate change vulnerability	Direct	1	Qualitative
Temporary scarcity	Direct	1	Qualitative
Risk of strategic use	Direct	1	Qualitative
Abundance in earth's crust	Direct	1	Ppm

(Frenzel et al., 2017)

Criticality classification of byproduct and carrier pairs



Develop and refine metrics to guide decision making



System	Supply elasticity 95% CI	Causes of inelasticity	
Zn-In	(-0.08, 0.29)	Supply limited by production capacity from carrier	Lack of global price setting mechanism
Cu-Se	(-0.03, 0.09)	Supply limit of carrier; Limit of recovery efficiency (~50%)	
Zn/Coal-Ge	(-0.31, 0.36)	National stockpiling strategy	

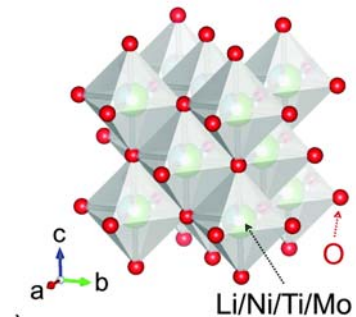
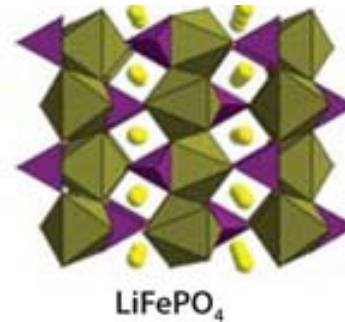
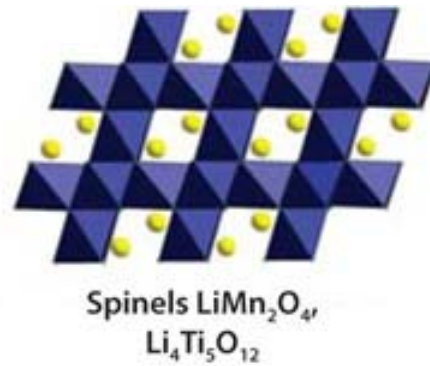
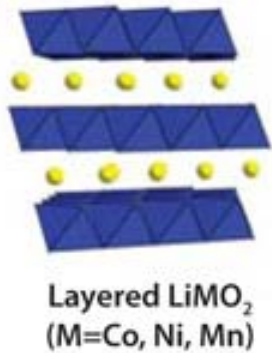
Byproduct status as indicator of criticality?

- Sometimes...
- Yes for indium
- Mixed evidence for selenium and germanium's inelastic supply, including:
 - the supply limit of carrier,
 - recovery efficiency limits,
 - lack of a global price-setting mechanism
 - national strategic stockpiling that disrupts market forces.
- Conclusion: Difference between supply and supply potential more indicative of criticality than 'byproduct dependence'



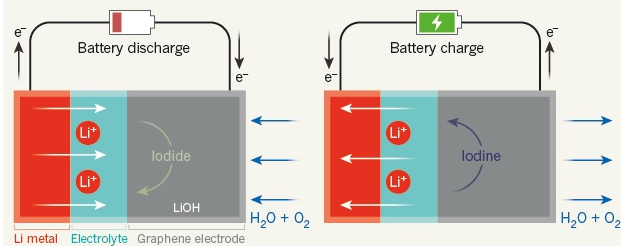
Novel materials development

Non-cobalt containing cathodes



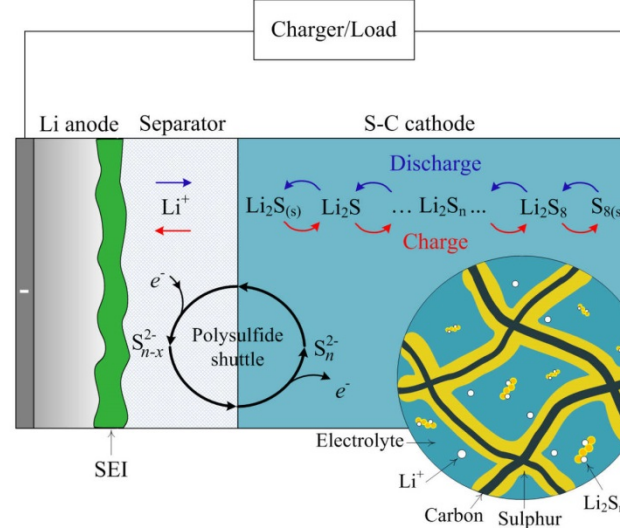
BREATHING BATTERIES

A new lithium-air battery shuttles lithium ions through an iodide-salt based electrolyte to react with oxygen at a graphene electrode.

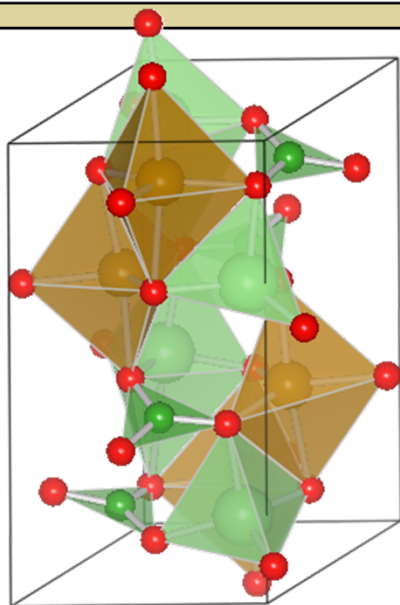


Li-air, Li-sulfur

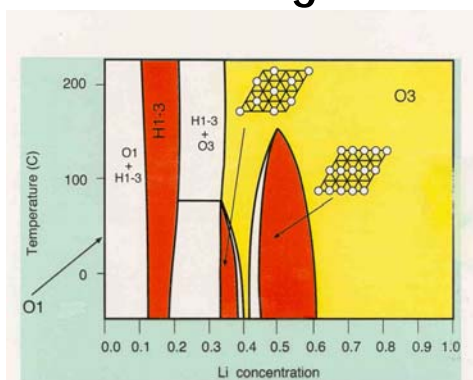
Solid state



Computational efforts have accelerated the materials discovery process



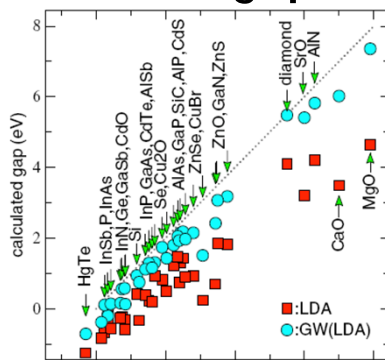
Phase diagrams



+



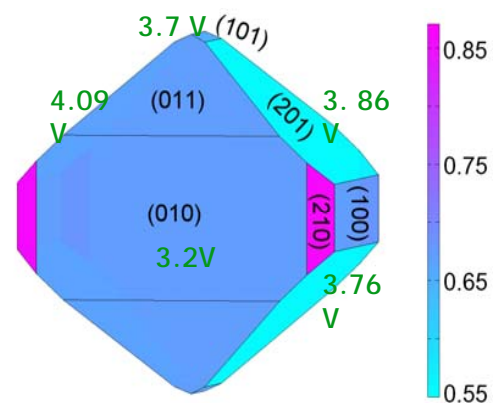
Bandgaps

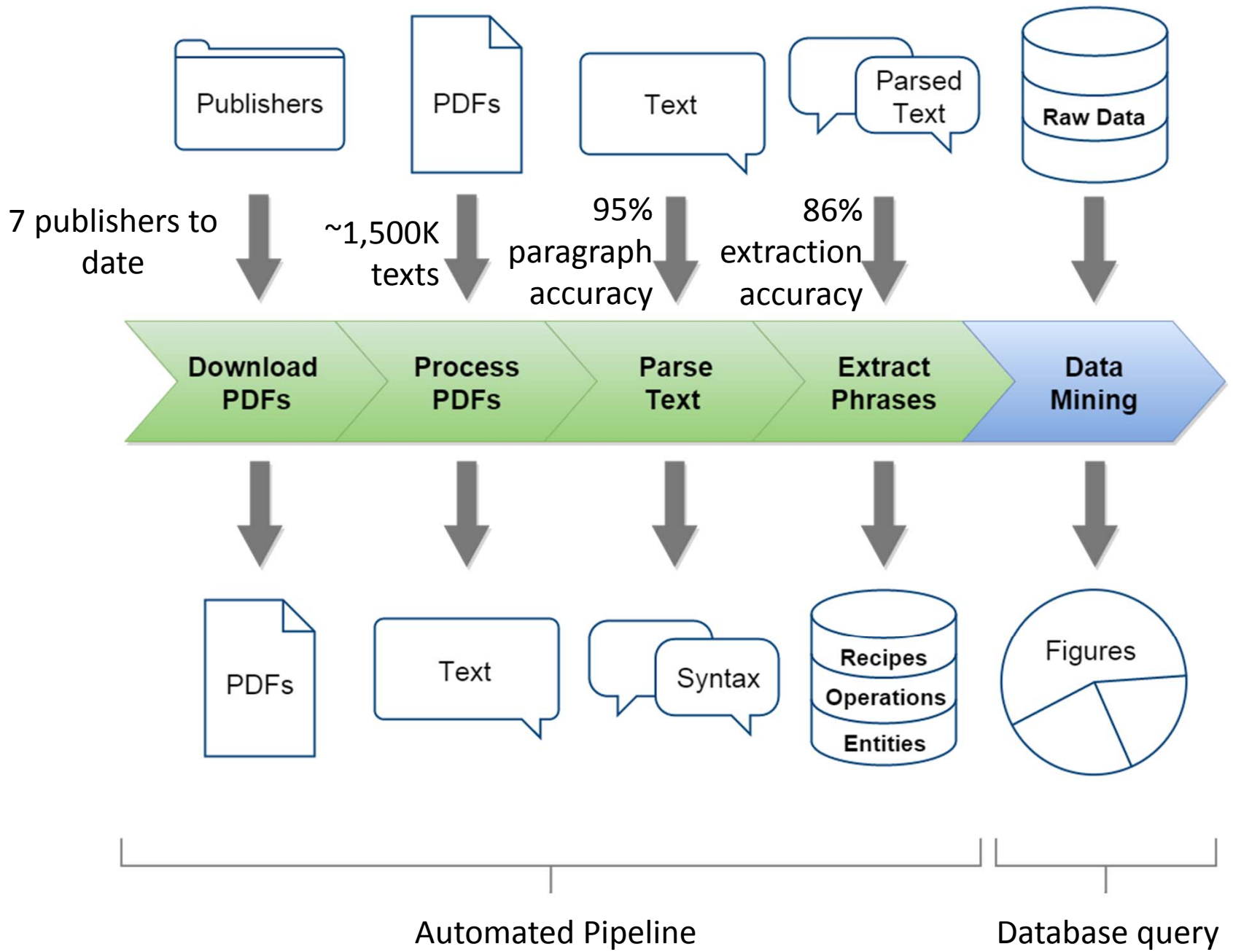


+

$$H = \sum_{i=1}^{N_e} \nabla_i^2 + \sum_{i=1}^{N_e} V_{nuclear}(r_i) + \frac{1}{2} \sum_i \sum_{j \neq i} \frac{1}{|r_j - r_i|}$$

Surfaces





Develop recipe database to improve understanding of materials synthesis



2. Experimental methods

NaNi_{1/3}Co_{1/3}Fe_{1/3}O₂ was synthesized by solid-state reaction. Excess amounts of Na₂O, NiO, Co₃O₄ and Fe₂O₃ were mixed and ball milled for 4 h at 500 rpm rate, and the resulting material was collected in the glove box. About 0.5 g of powder was fired at 800 °C under O₂ for 14 h before it was quenched to room temperature and moved to a glove box filled with argon.

X-ray diffraction (XRD) patterns were collected on a PANalytical X'Pert Pro equipped with Cu Kα radiation in the 2θ range of 5–85°.

Identify hundreds of thousands of manuscripts by target material

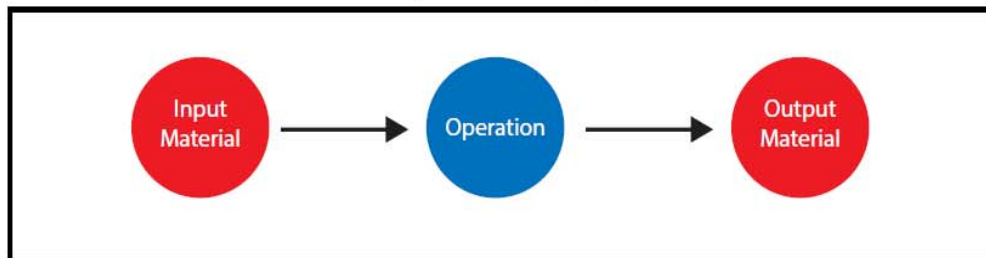
Generate codified, machine readable database of recipes

Extract synthesis text through machine learning and rule-based methods

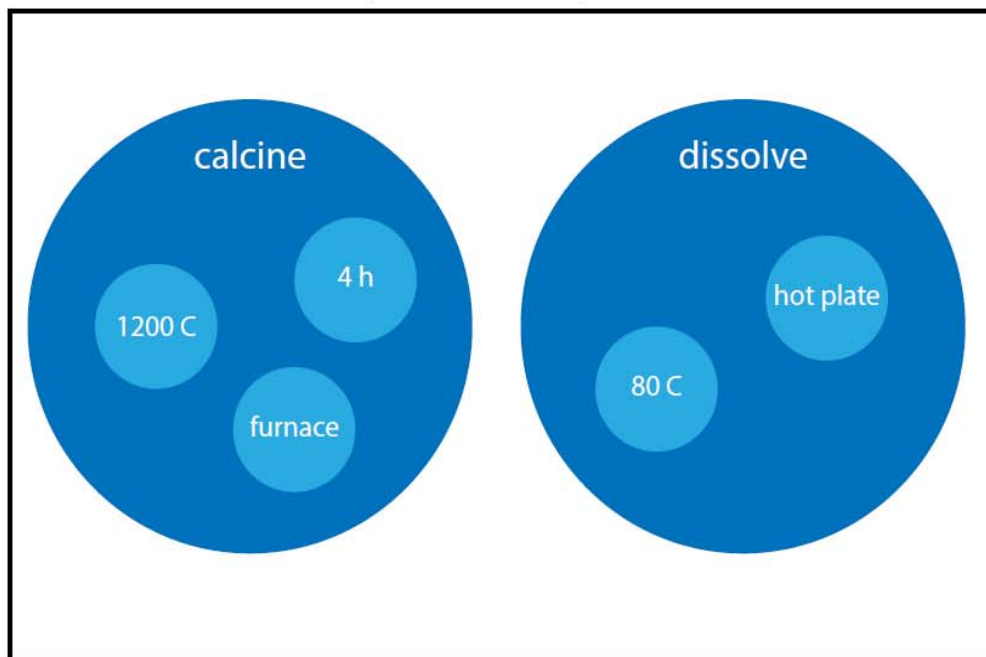
NaNi_{1/3}Co_{1/3}Fe_{1/3}O₂ was synthesized by solid-state reaction. Excess amounts of Na₂O, NiO, Co₃O₄ and Fe₂O₃ were mixed and ball milled for 4 h at 500 rpm rate, and the resulting material was collected in the glove box. About 0.5 g of powder was fired at 800 °C under O₂ for 14 h before it was quenched to room temperature and moved to a glove box filled with argon.



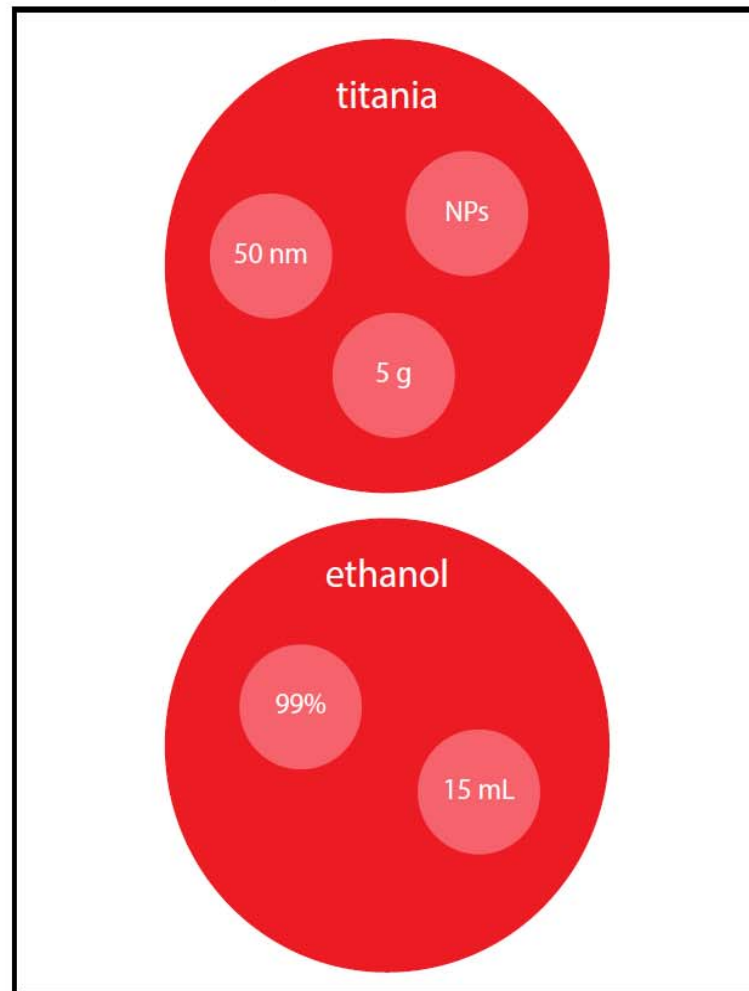
Extracted Synthesis Recipe Schema



Example Extracted Operations

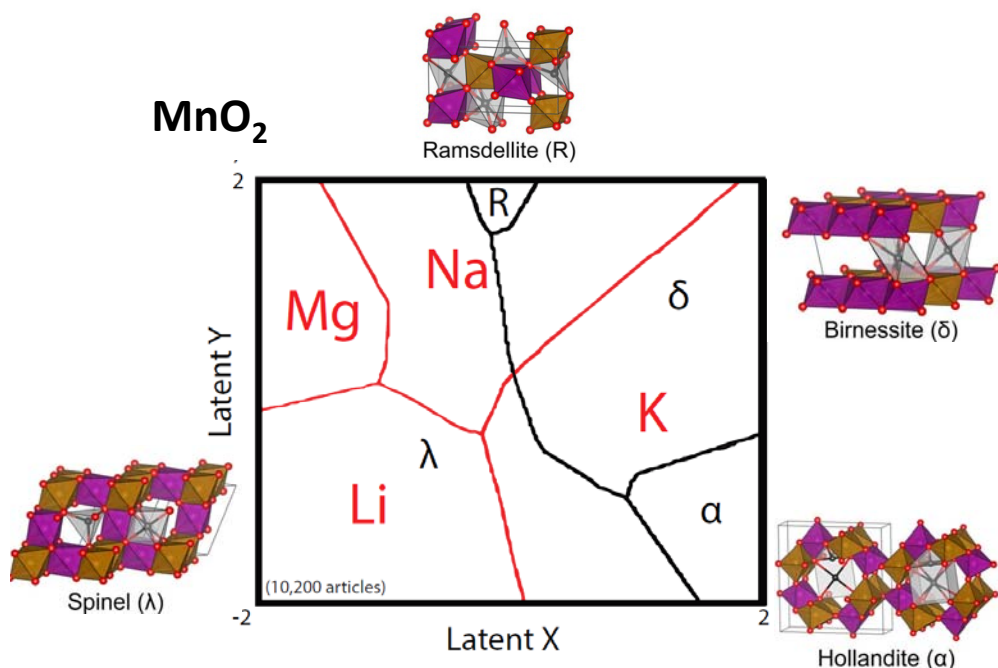


Example Extracted Materials



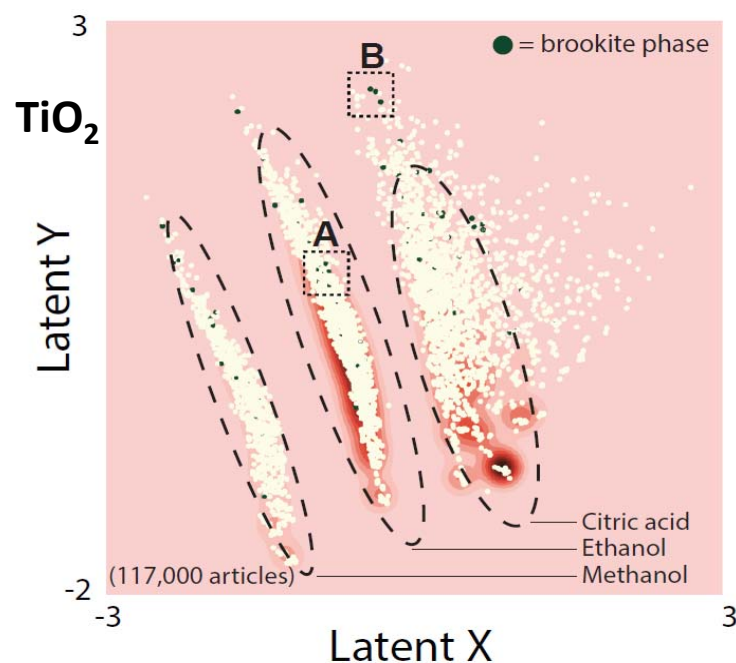
Resource-effective material development: suggesting multiple routes for synthesis

Suggesting synthesis conditions for materials, extend to novel materials

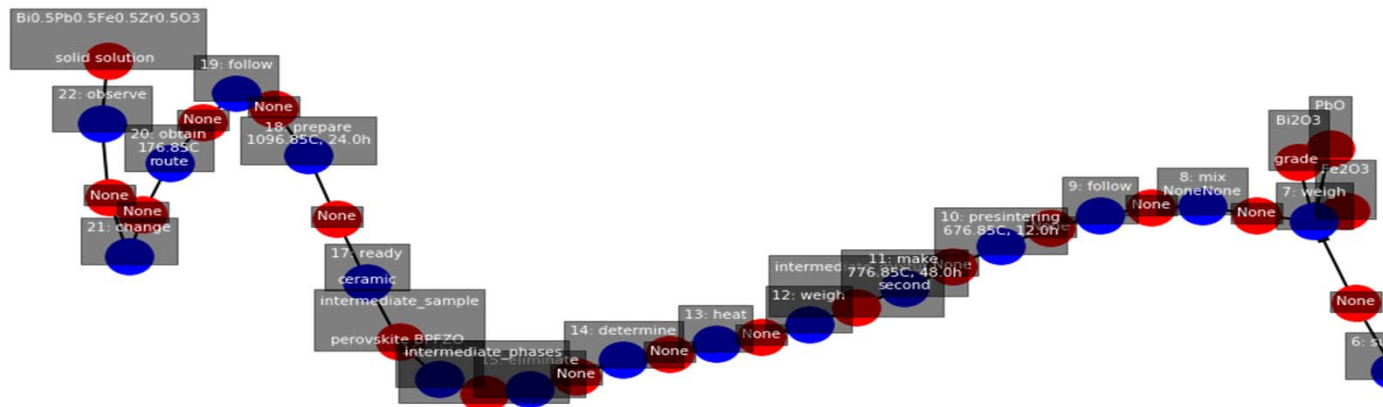
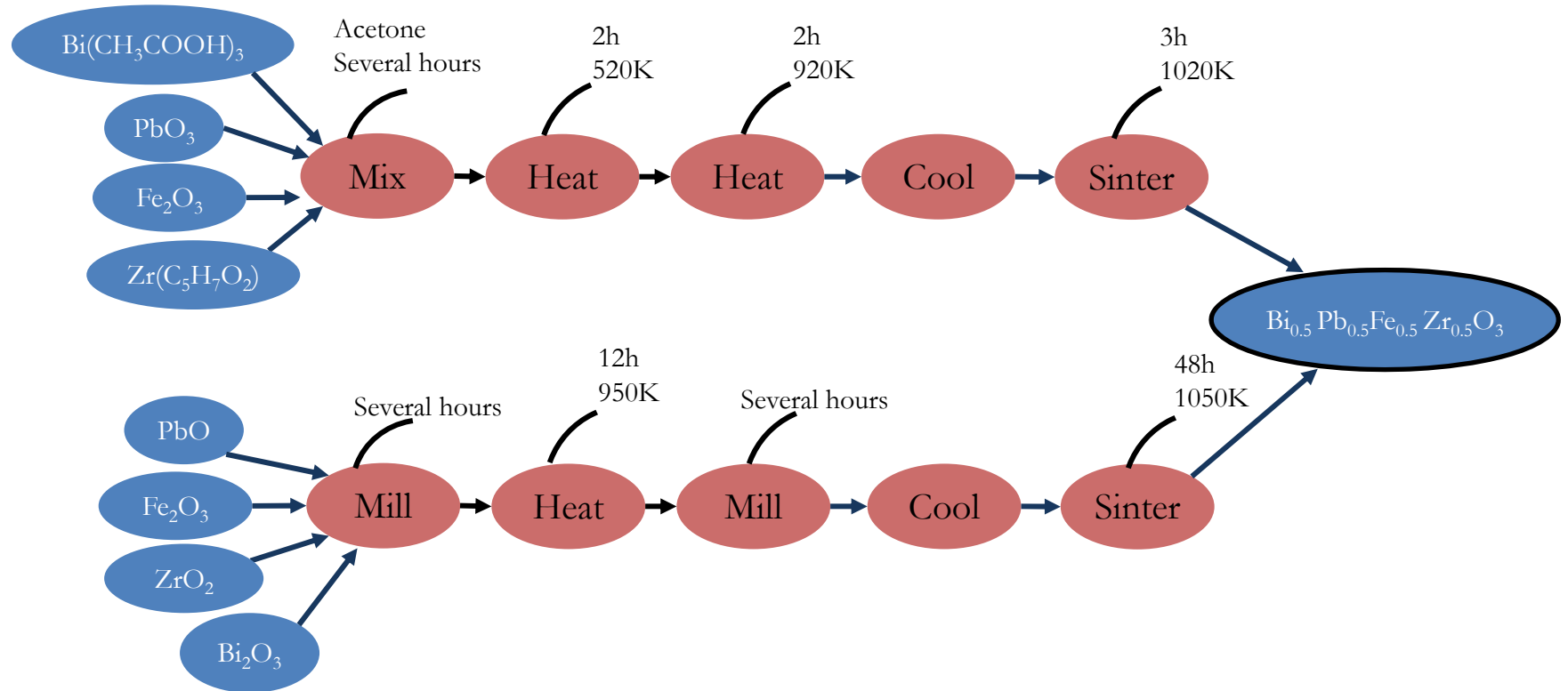


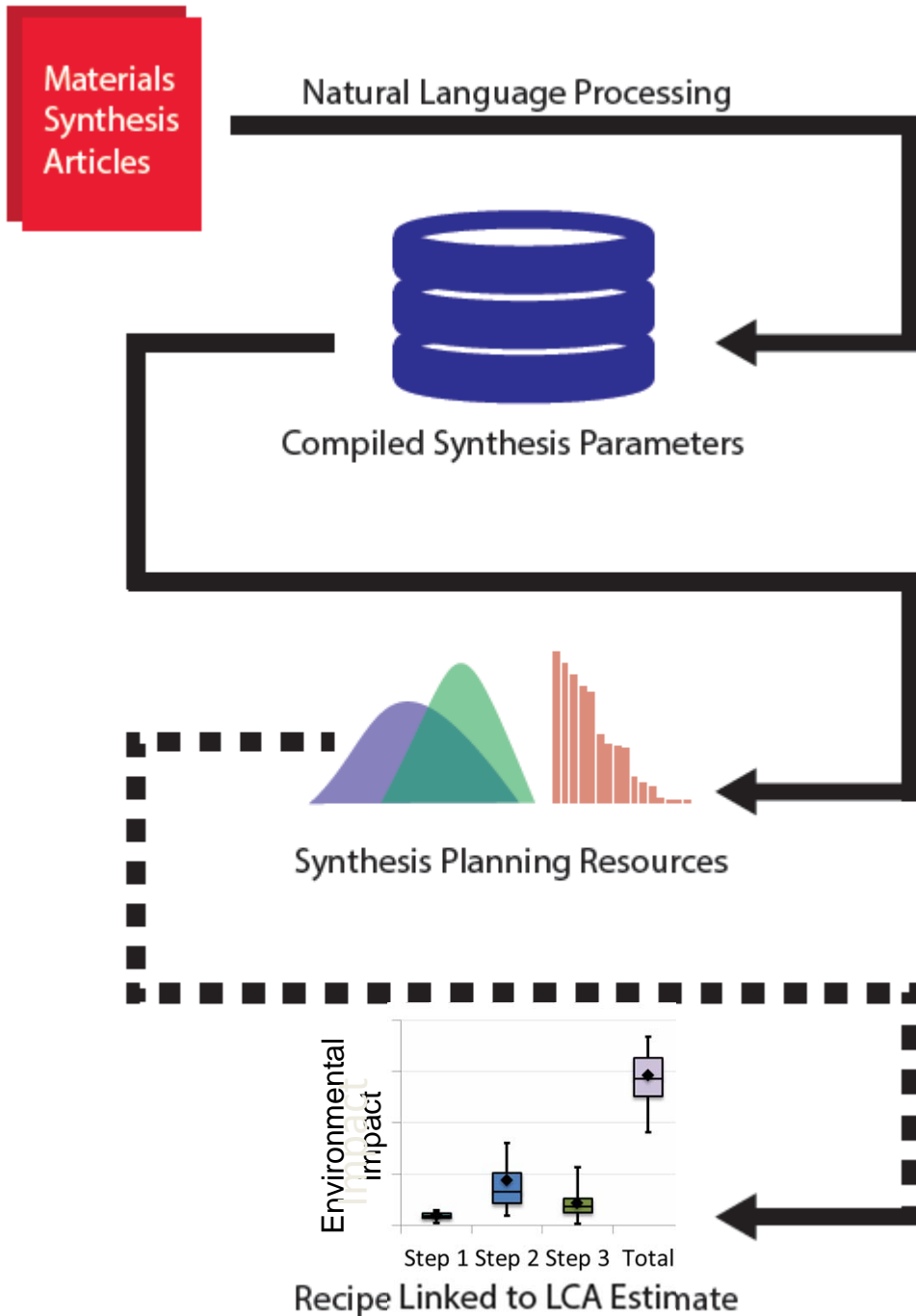
Polymorphs for MnO₂ overlaid with most probable alkali-ion use in synthesis (intercalation-based phase stability)

Clustering of latent space points to role for NaOH and ethanol in brookite synthesis



Generate recipes





Resource-effective performance of materials

- **Systems thinking divorced from materials science and vice versa masks opportunities**
 - As system, material, and process complexity increases need novel ways to analyze problems
- **Materials have critical role in solving key economic and environmental problems**
 - Solutions that account for scale are needed to address the impact of materials use



Thank you

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Making Matter Matter

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MIT Center for Materials Science and Engineering

NSF

MITei

Office of Naval Research
ONR
Science & Technology

intel

DELL

TATA CENTER
TECHNOLOGY + DESIGN

lenovo

HYDRO

MIT Portugal

hp

AMD

American Forest & Paper Association

LORD

MATERIALS PROJECT

Collaborators:
Prof. Gerbrand Ceder,
Prof. Andrew Mccallum
Prof. Stefanie Jegelka
Michael Larcy,
Dr. Thomas Pointot,
Prof. John Ochsendorf,
Prof. Hamlin Jennings

Thank you
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