


# Harnessing High Temperature Materials for Extraction and Processing



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Department of Materials Science & Engineering

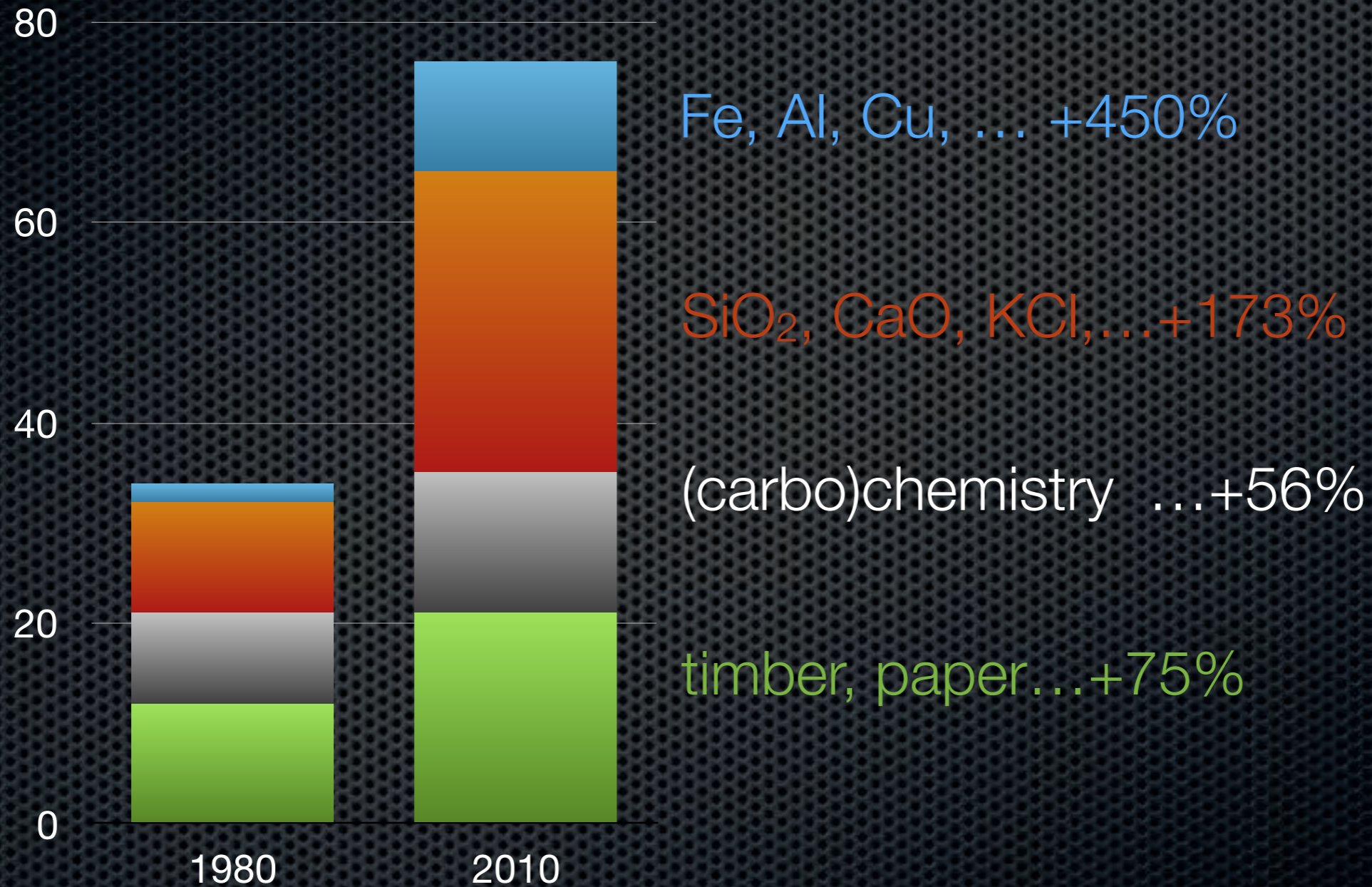
Massachusetts Institute of Technology, USA

MIT ILP R&D Conference, November 15<sup>th</sup>, 2017

Cambridge, MA

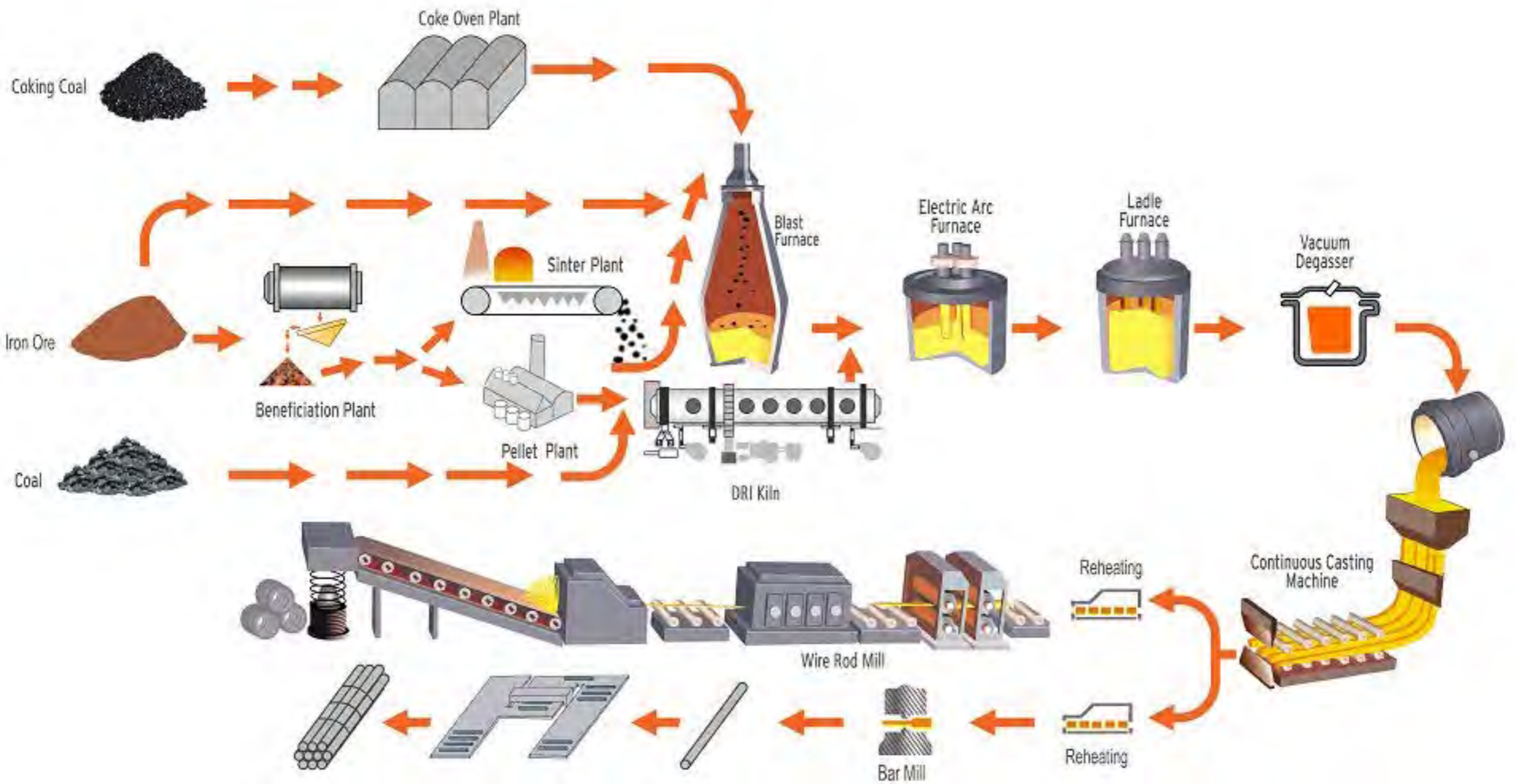
# Materials for 9 billion people?

**billion tonnes** (data: SERI2012)



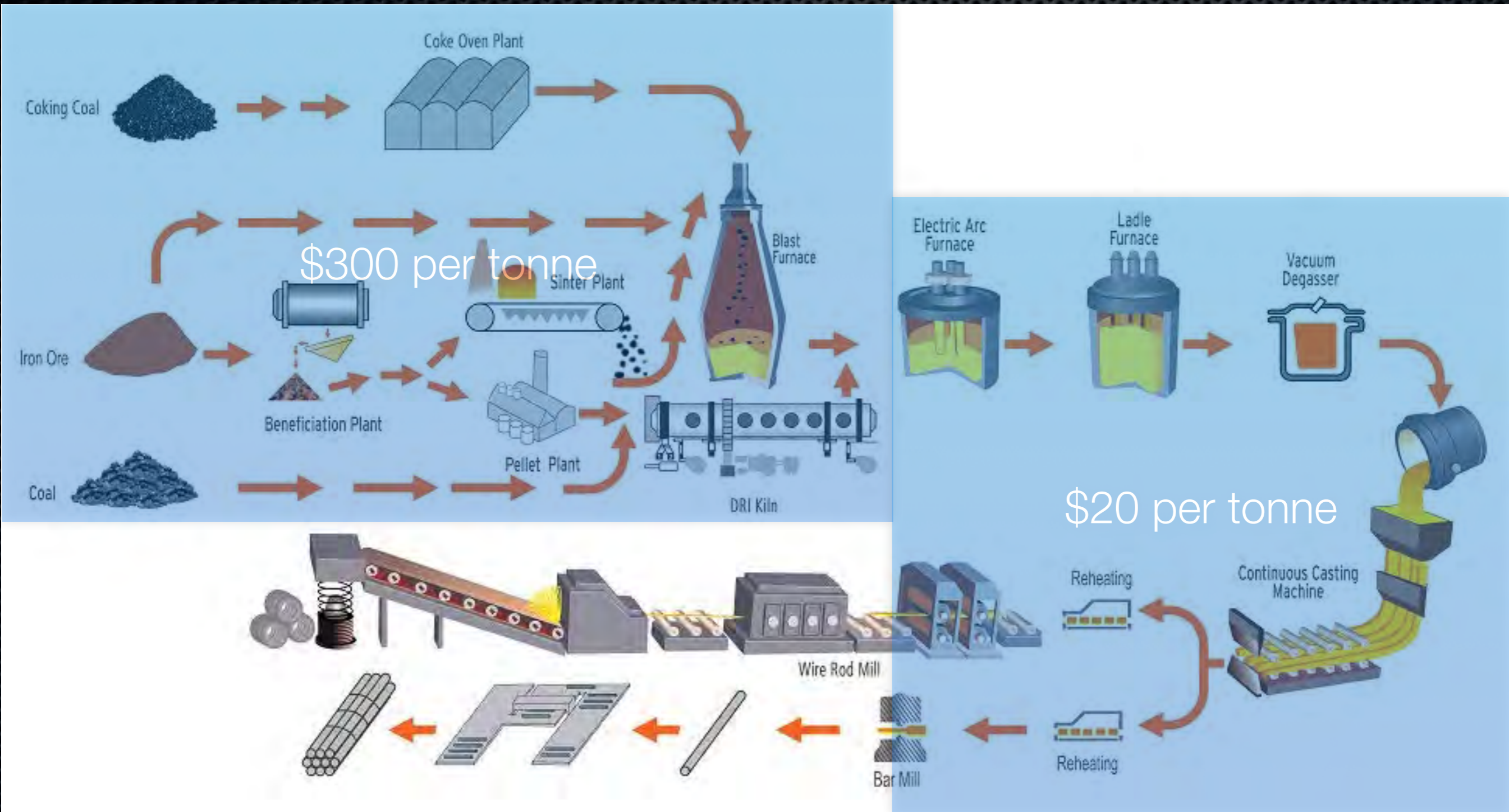
fastest growth is in minerals and metals

# Metals processing: steel



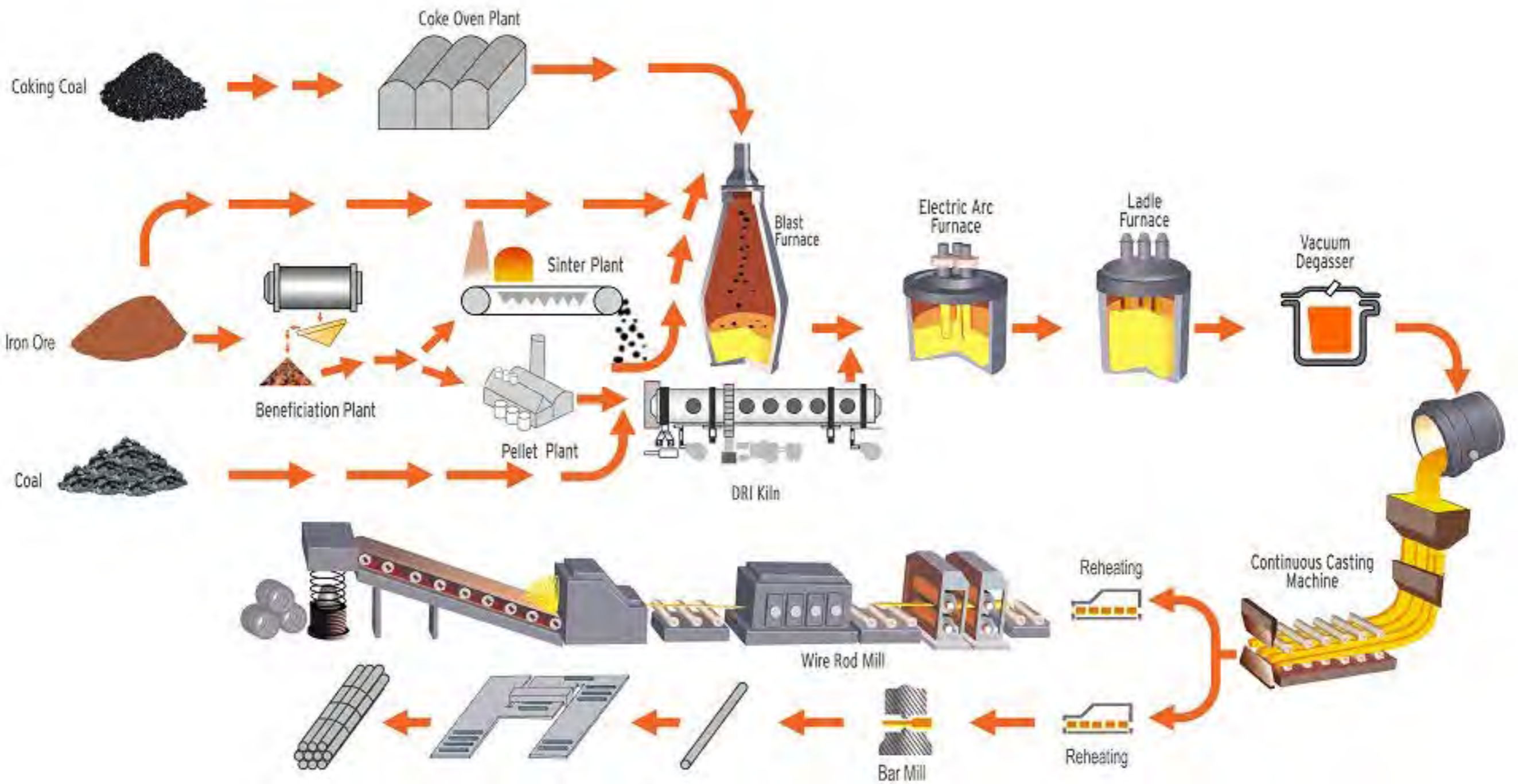
typical plant capacity: 114 tonnes each 1/2 hour

# Metals processing: steel



cost < \$0.32 per kg (less than retail price of flour)

# Metals processing: steel



the liquid state and high temperature are ubiquitous

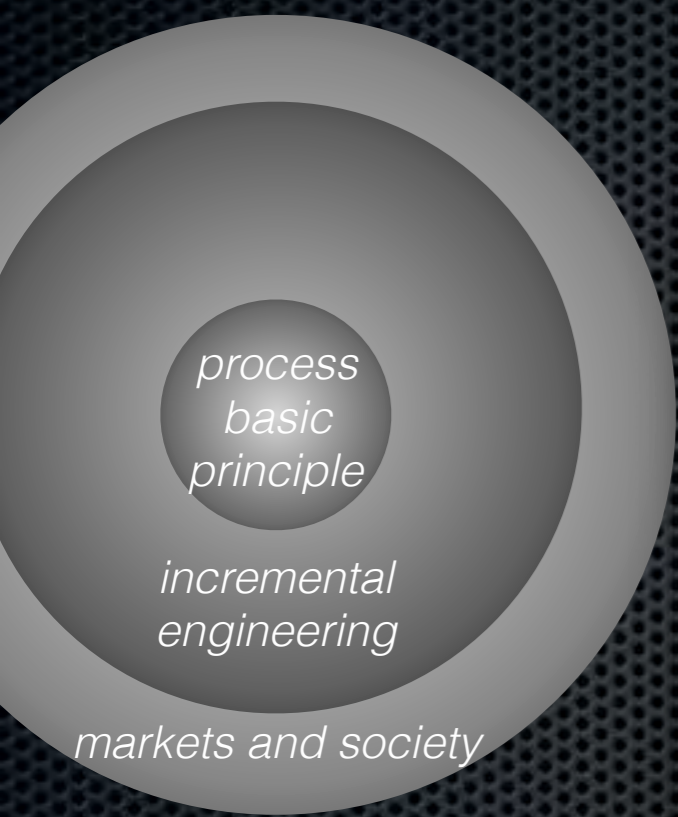
# Metals extraction involves melts

Copper - 20 Mt per year  
3600 kWh / t  
from concentrate to  
copper cathodes  
conversion costs: \$320/t

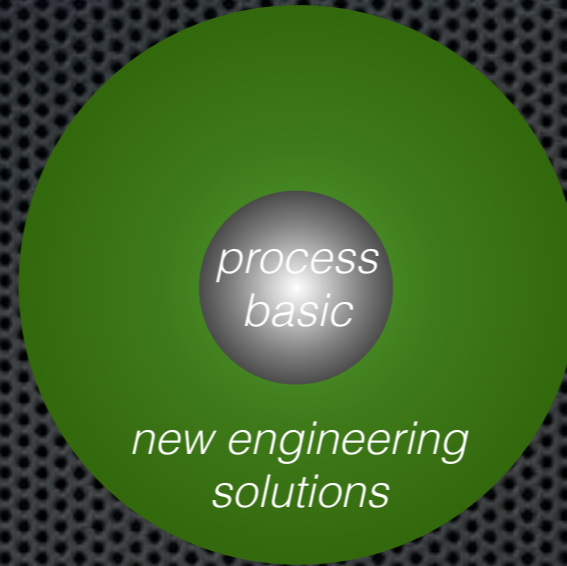
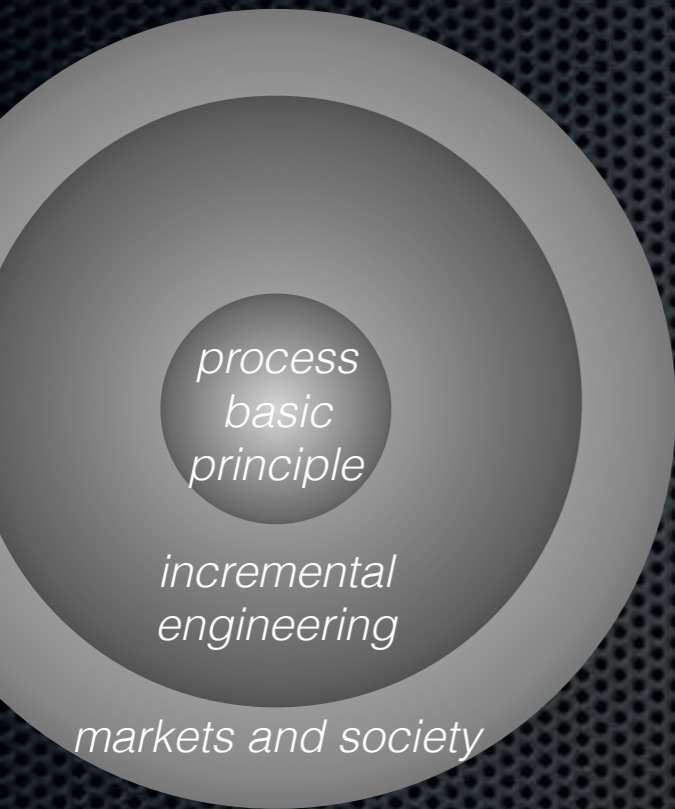


Aluminium - 60 Mt per year  
electrolysis  
conversion costs: \$1000/t

Existing  
processes



Existing  
processes



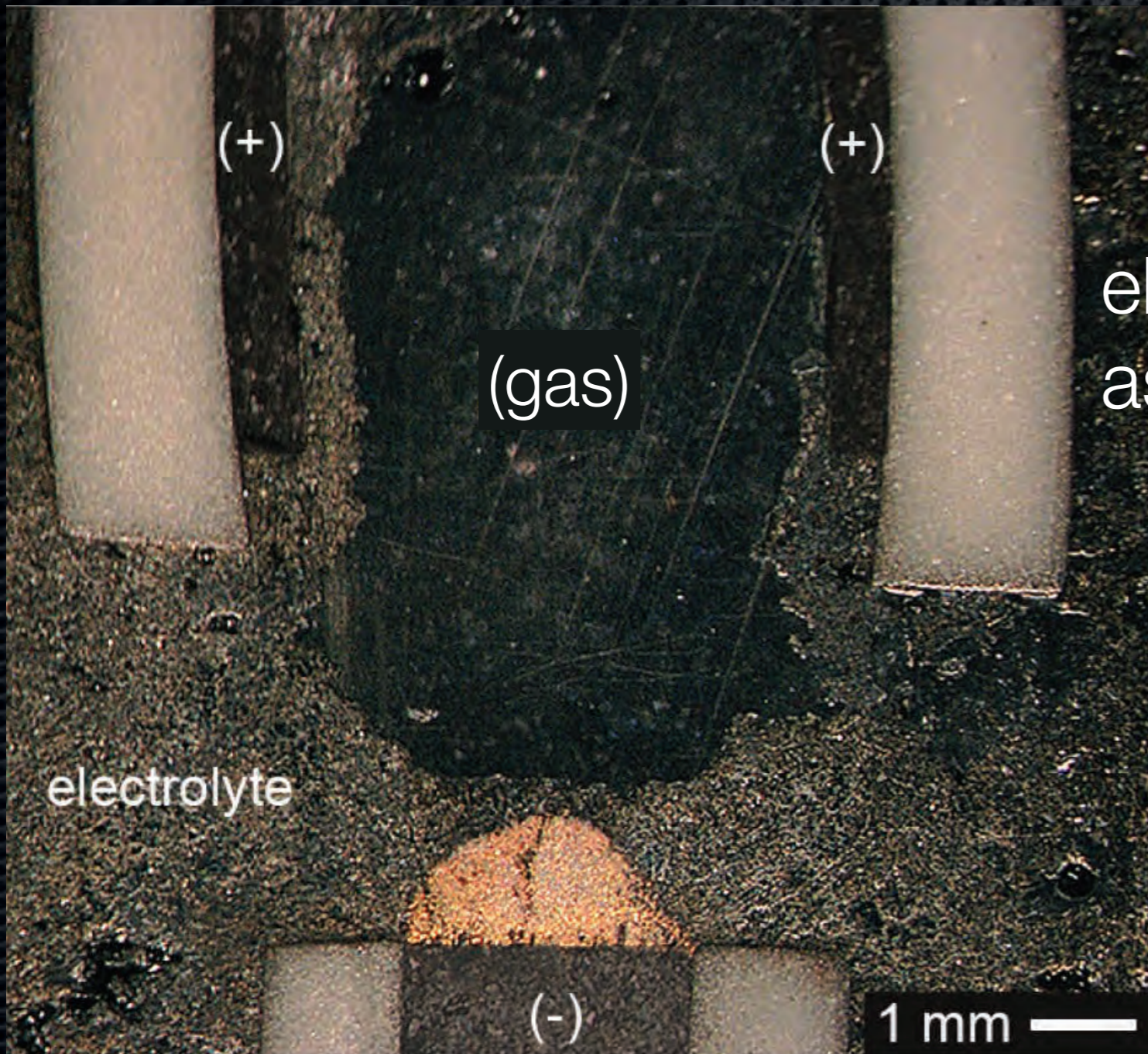
higher  
productivity



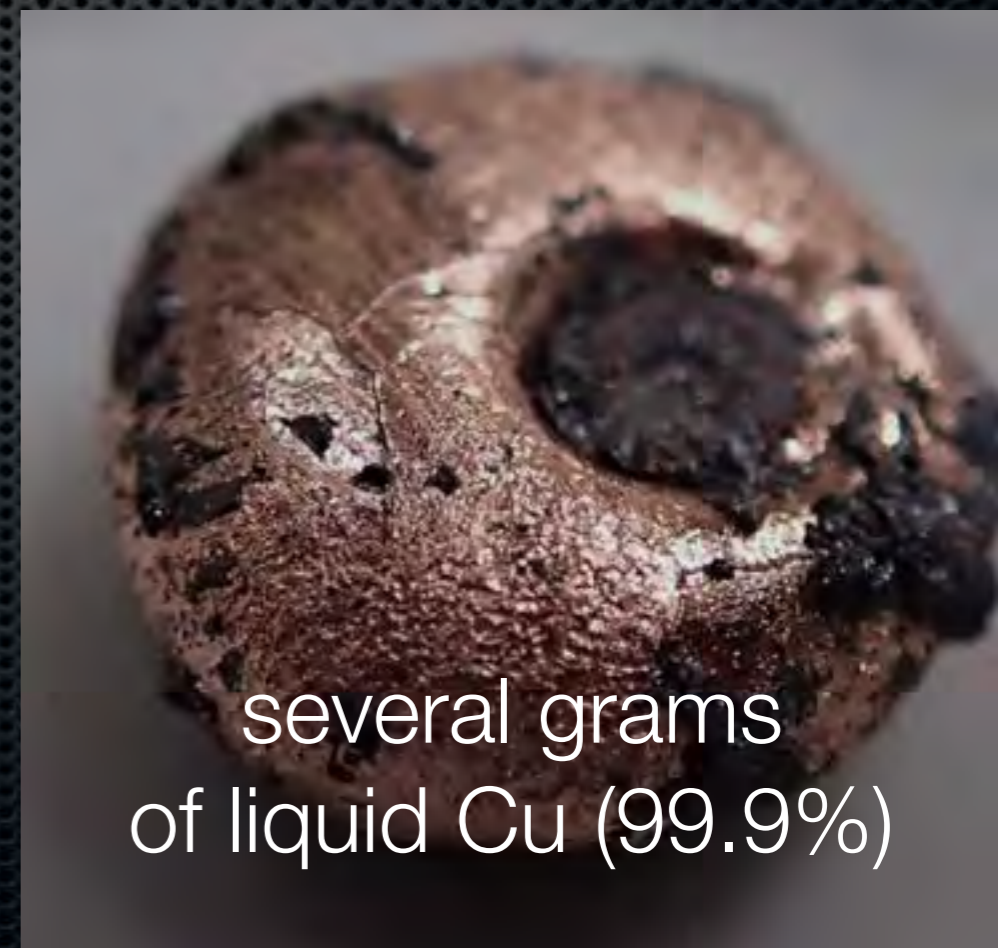
higher  
selectivity



# Sulfides electrolysis

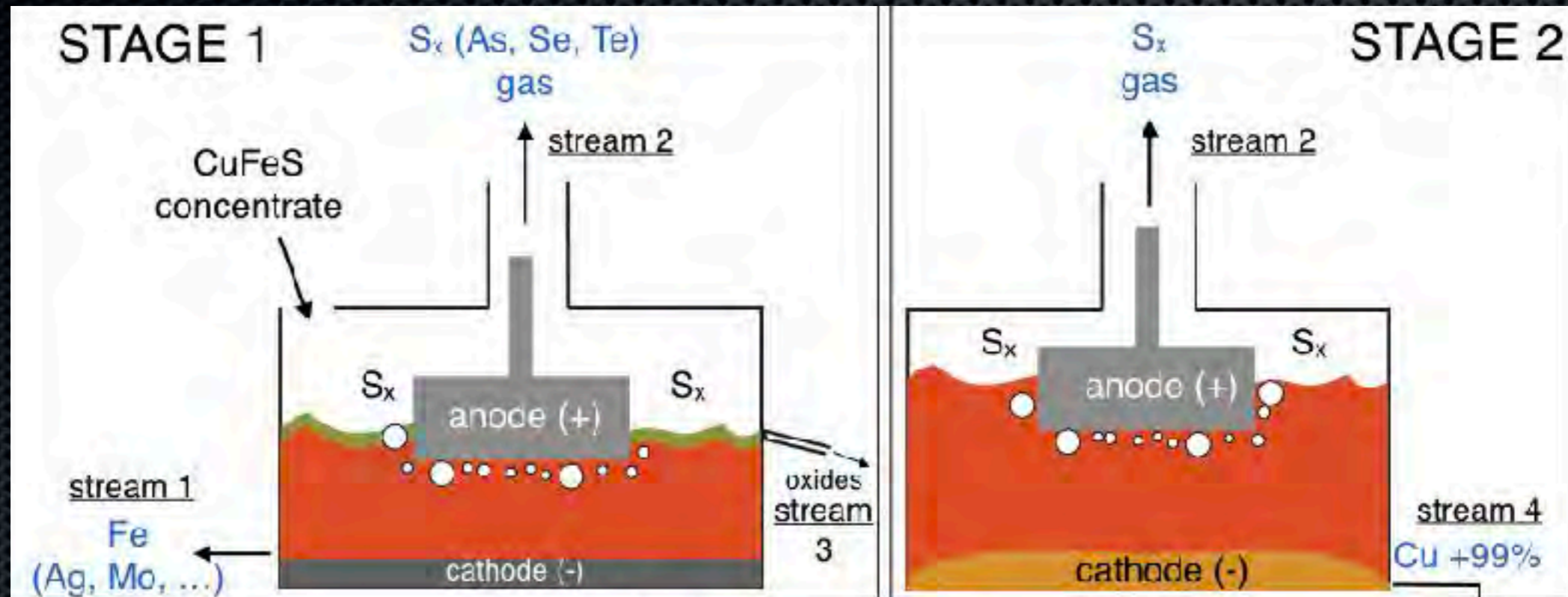


elemental sulfur  
as anodic product



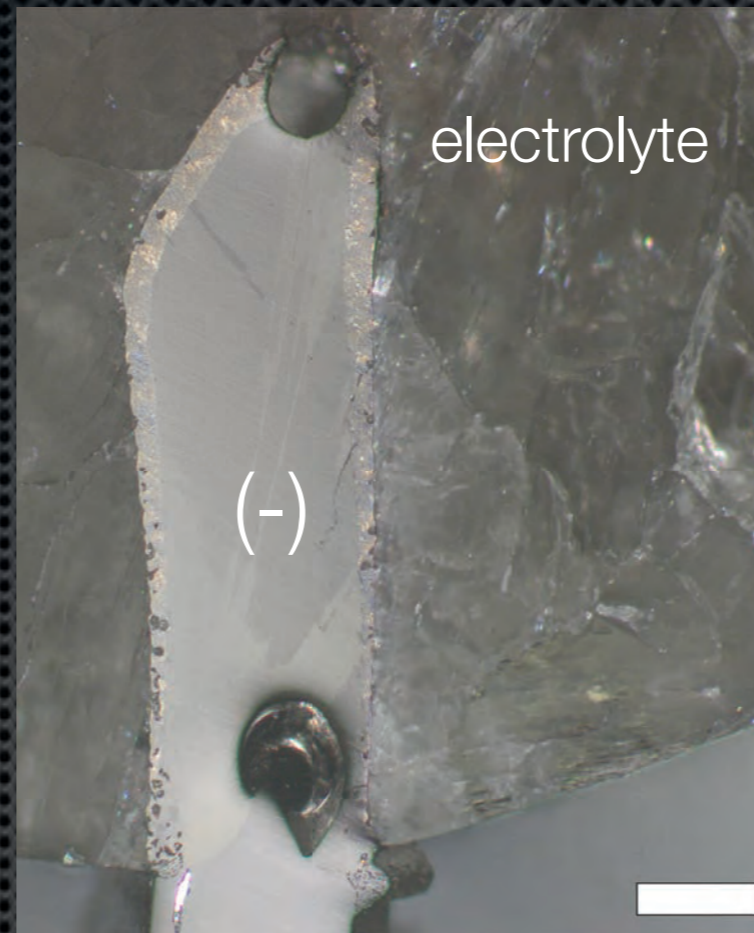
electrolytic decomposition of metal sulfides

# Sulfides electrolysis



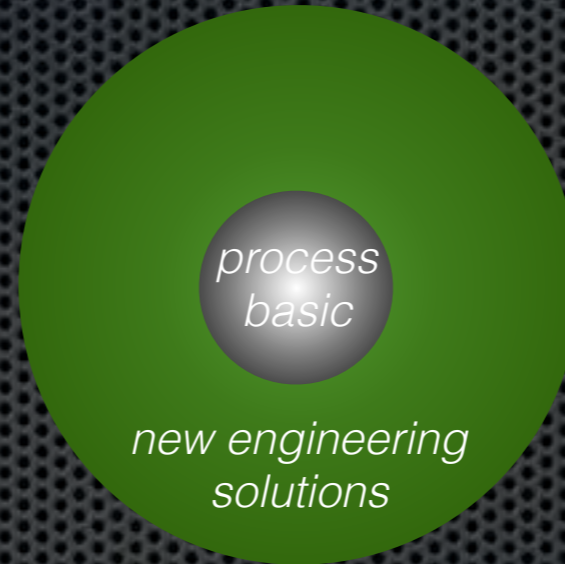
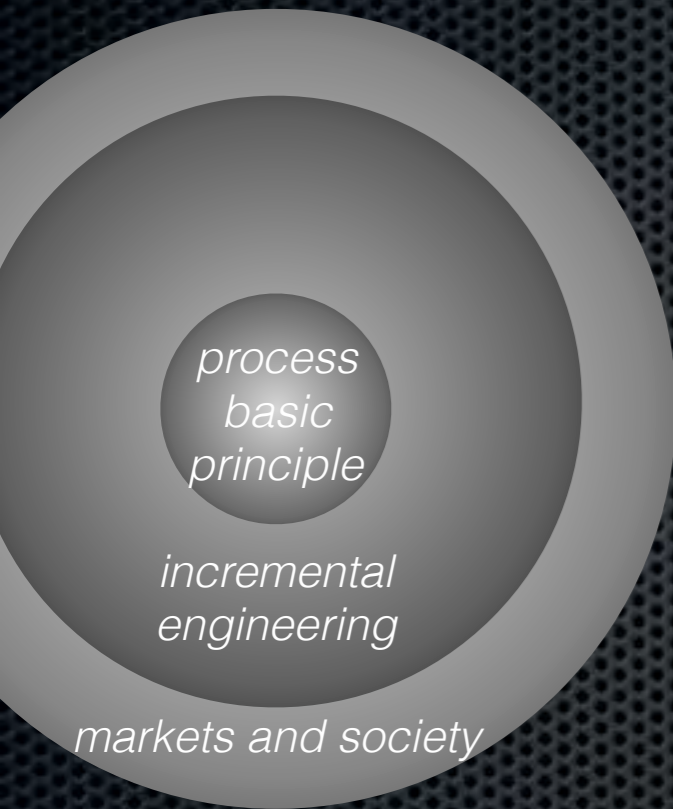
- productivity 6x the one of aluminium electrolysis
- electrical energy consumption around 3000kWh/t (-20%)
- sulfur as by-product with selective recovery of As, Se, Te
- principles transferable to Mo, Ag, Au, Zn, Co, Ni ...

# Molten rare-earth oxides electrolysis



- high selectivity for La vs Y (99 vs 1)
- compatible with rare-earth oxide concentrates
- enables to produce refractory alloys (e.g. Ir-La)
- production of oxygen as by-product

Existing  
processes



higher  
productivity

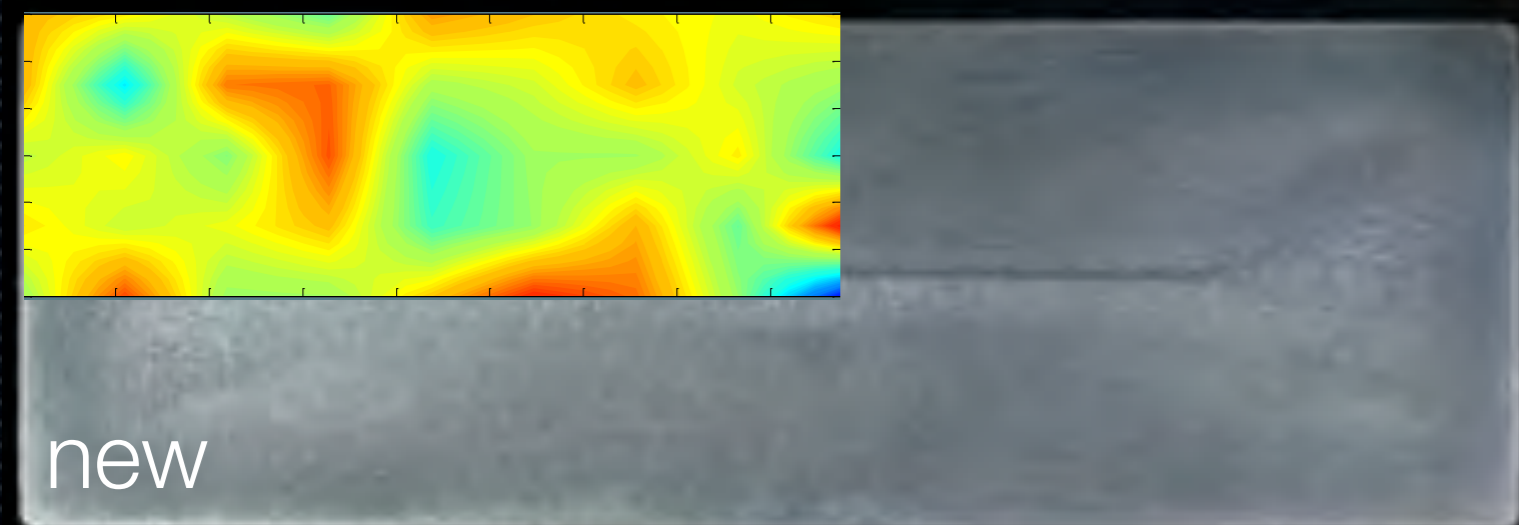
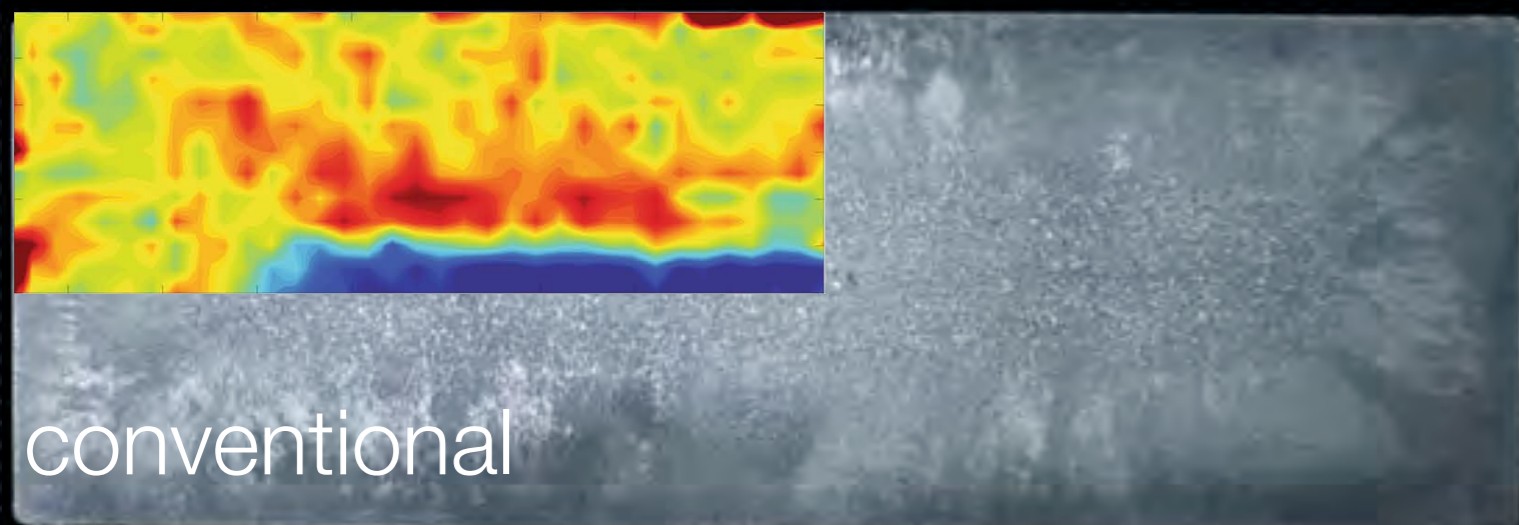


higher  
selectivity

Multi-physics of aluminium casting for more recycling

Direct-chill casting of aluminium slabs

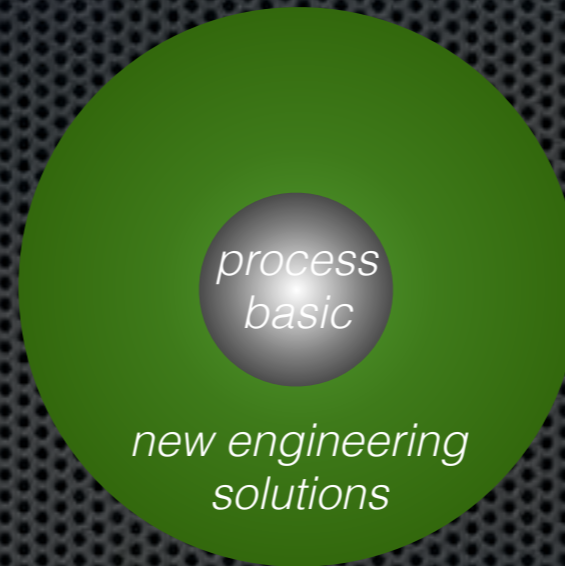
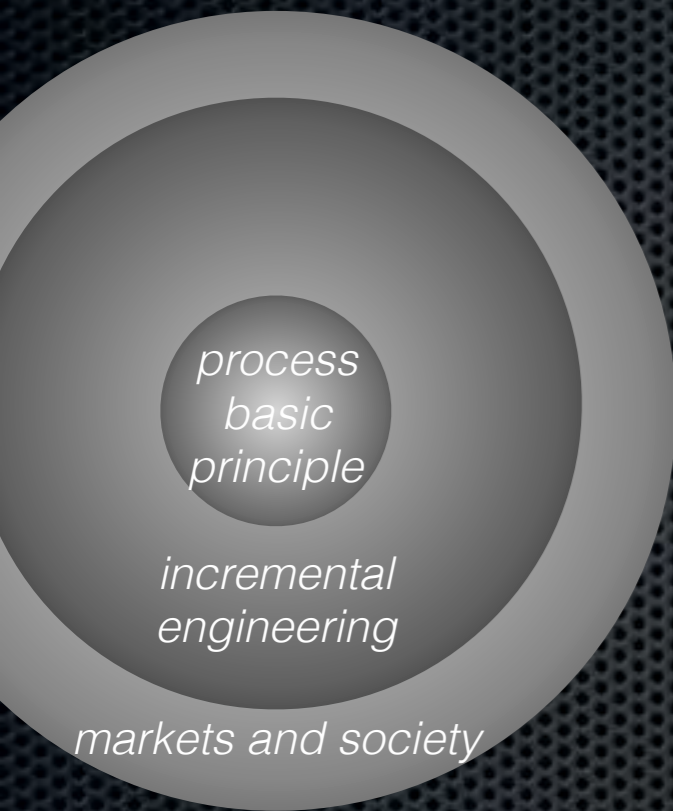




← 1.75 m →

- demonstration of the role of 2-phase flow in DC casting
- novel, cost-effective method to improve alloy elements distribution
- 100 tonnes of Al products, rolled into +10 miles of sheet
- 20% increase in productivity at the plant

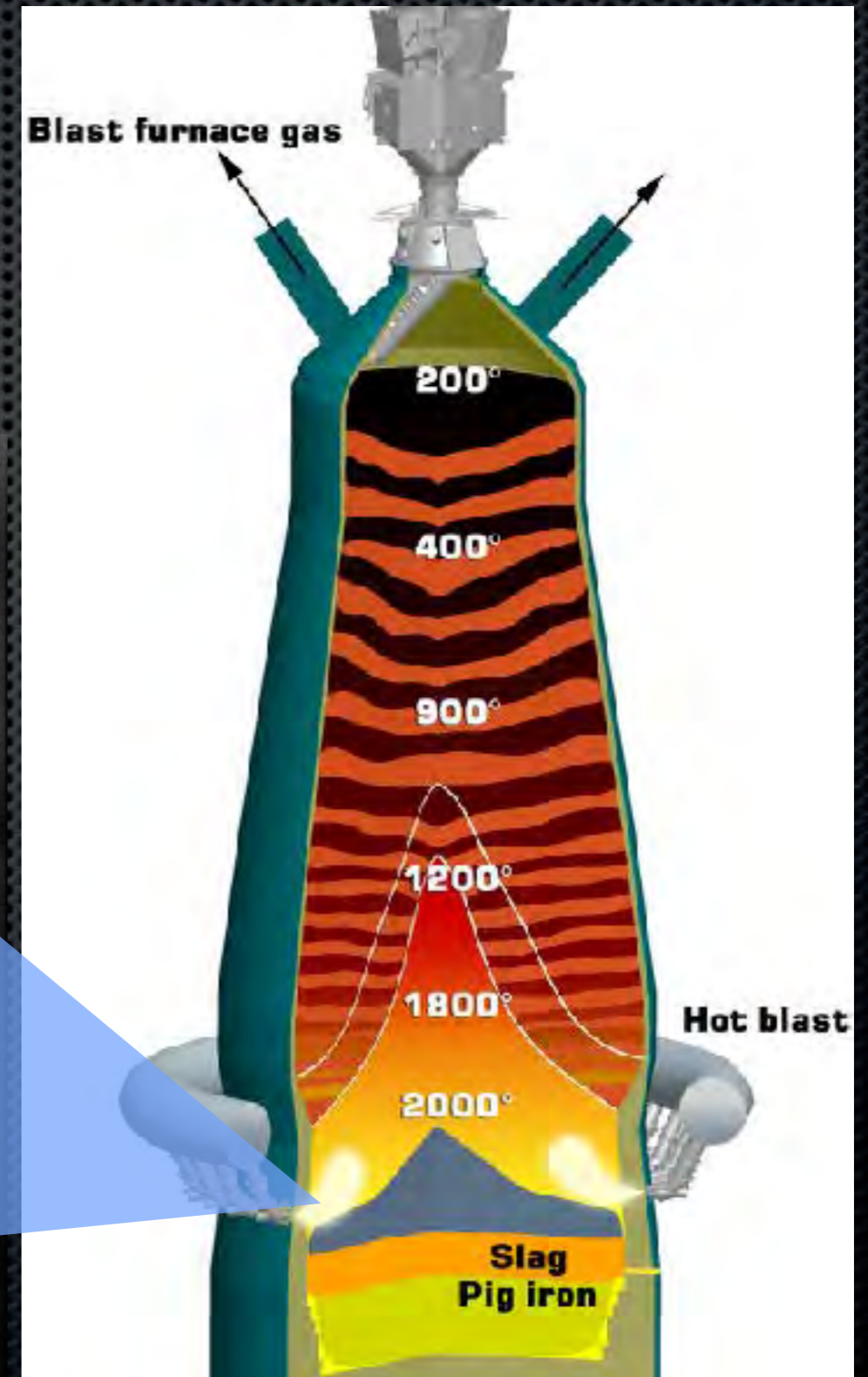
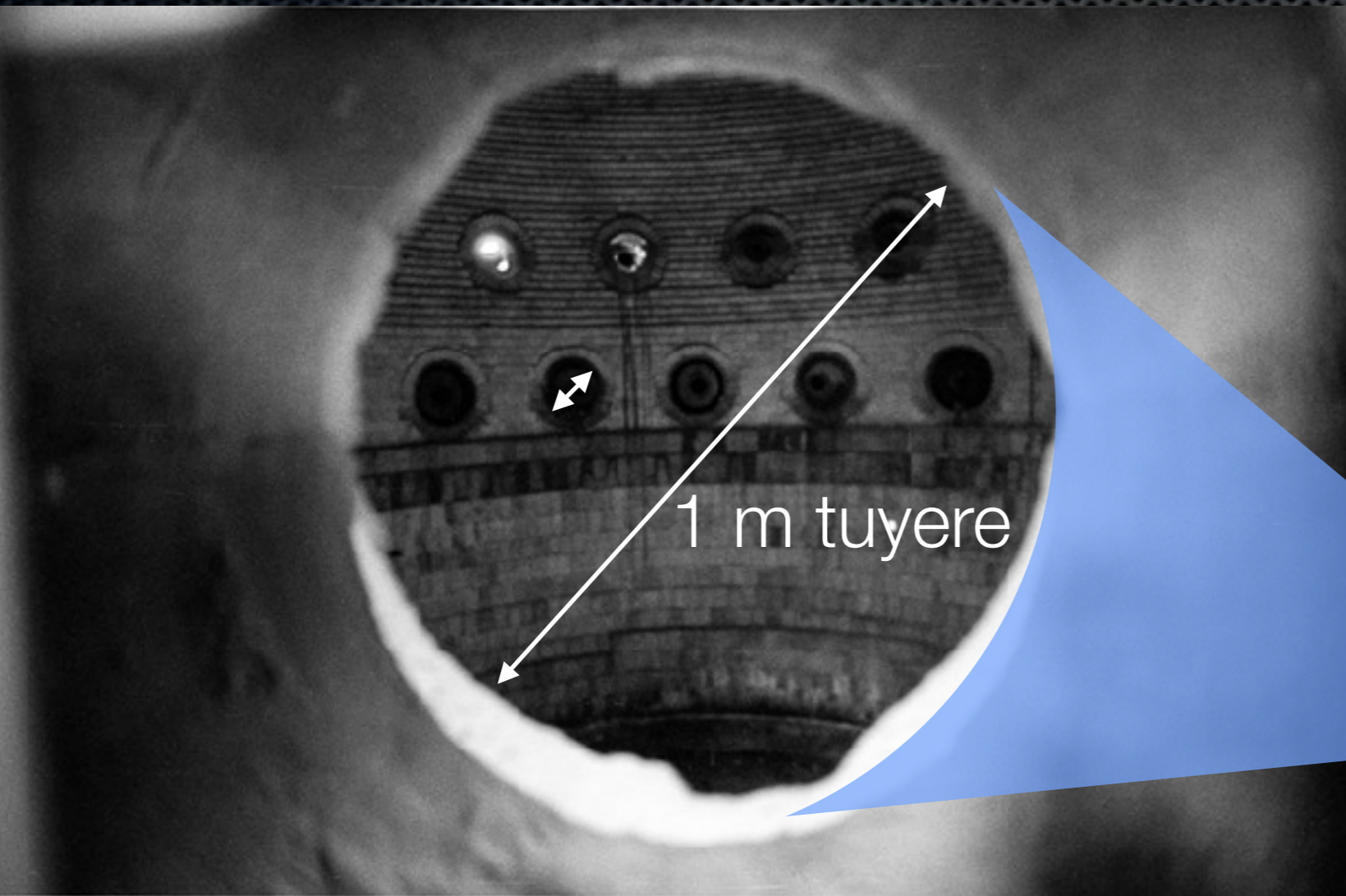
Existing  
processes



higher  
productivity

how about energy efficiency improvement?

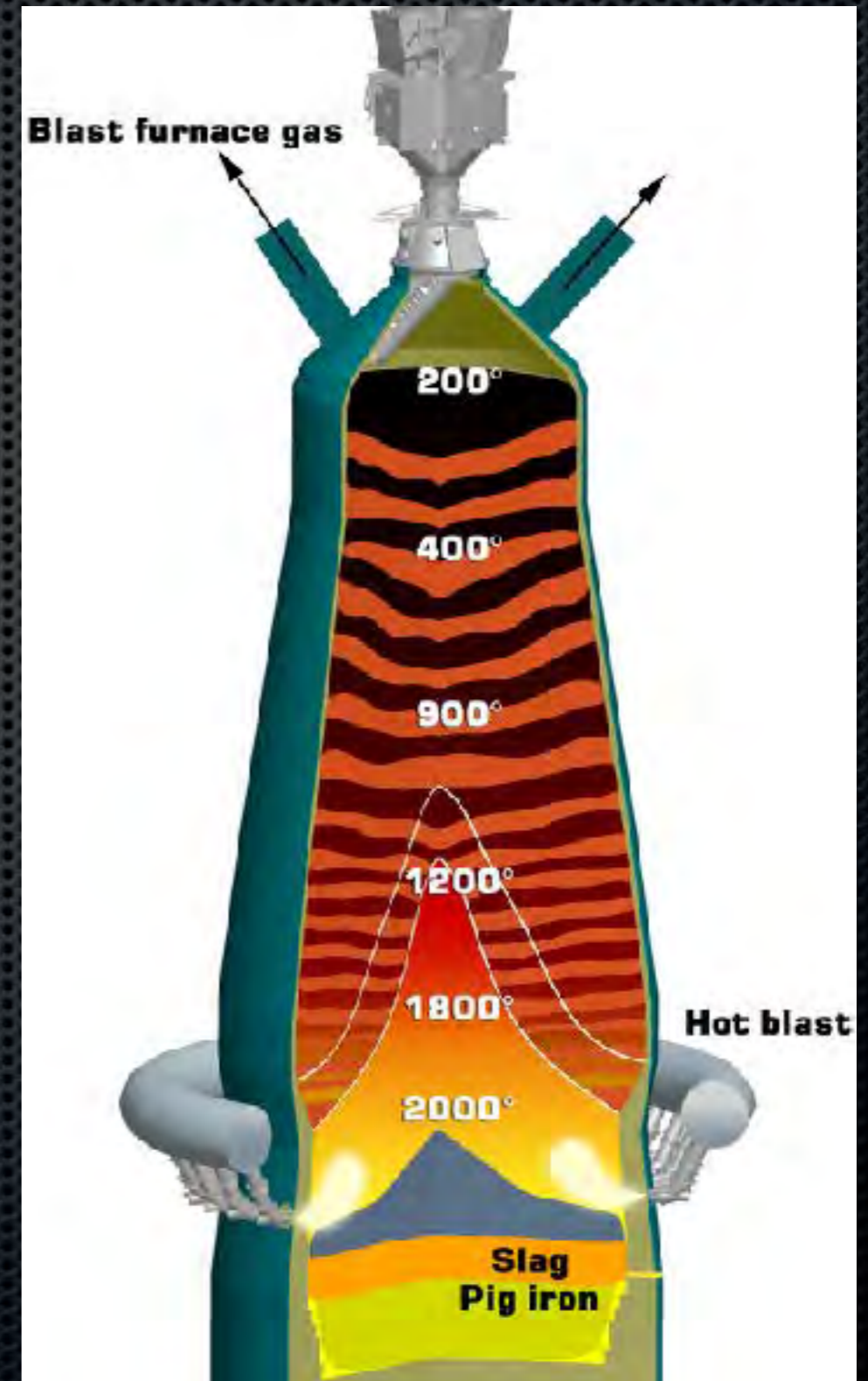
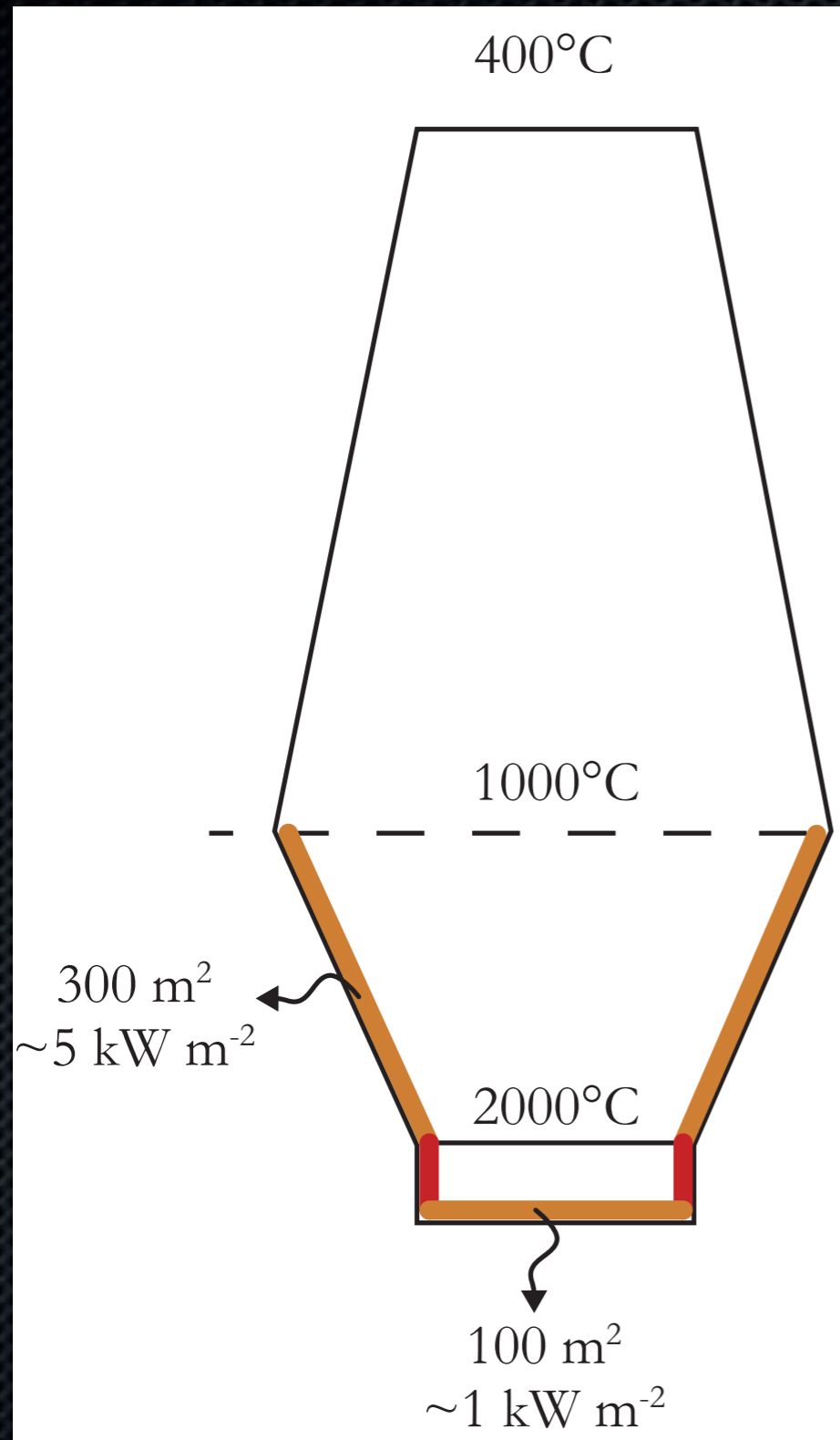
# Heat management : steel



Blast furnace,  $H=30\text{m}$ ,  $\varnothing=15\text{m}$



# Heat management : steel



Blast furnace,  $H=30\text{m}$ ,  $\varnothing=15\text{m}$

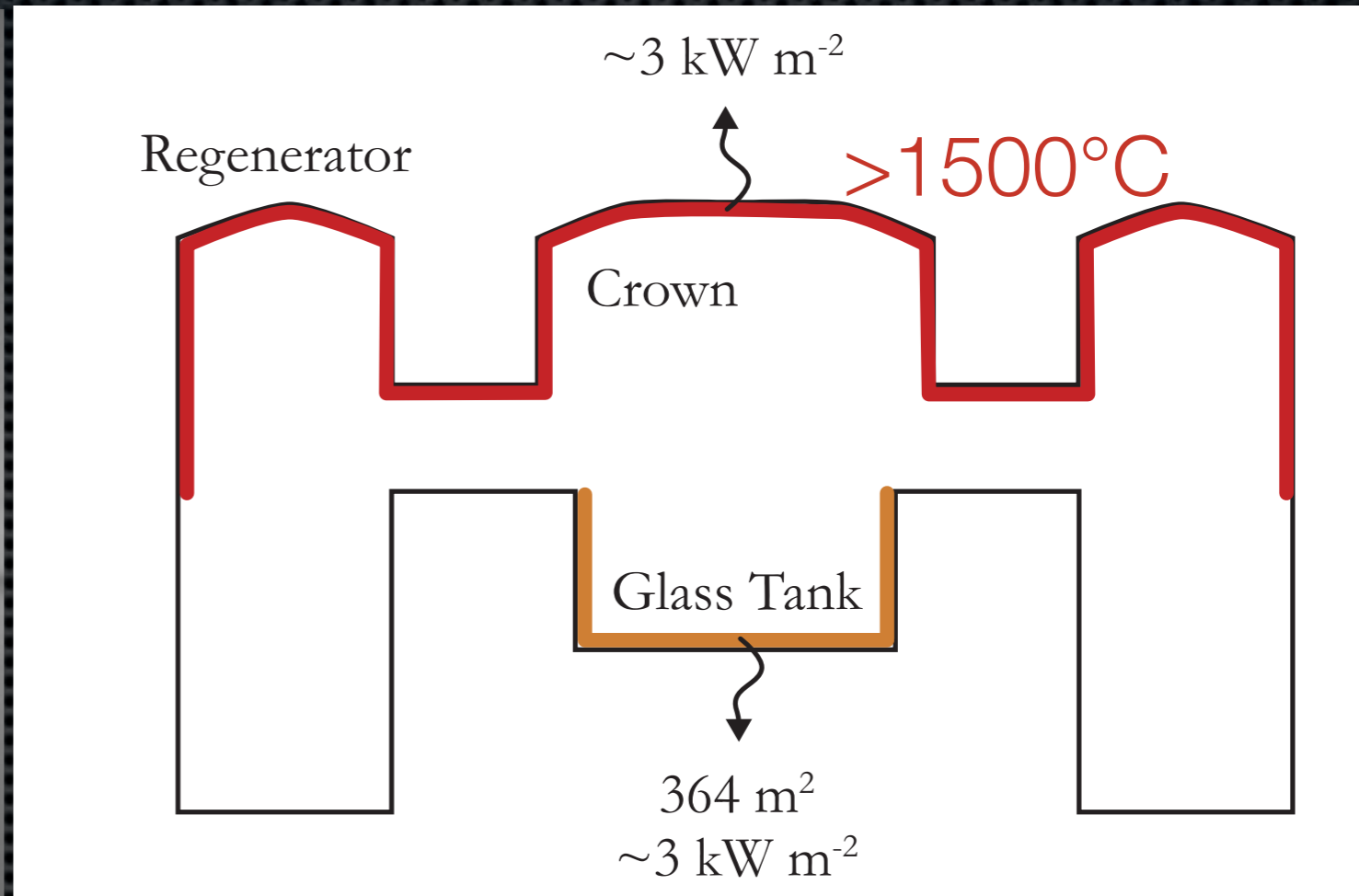
# Heat management : glass



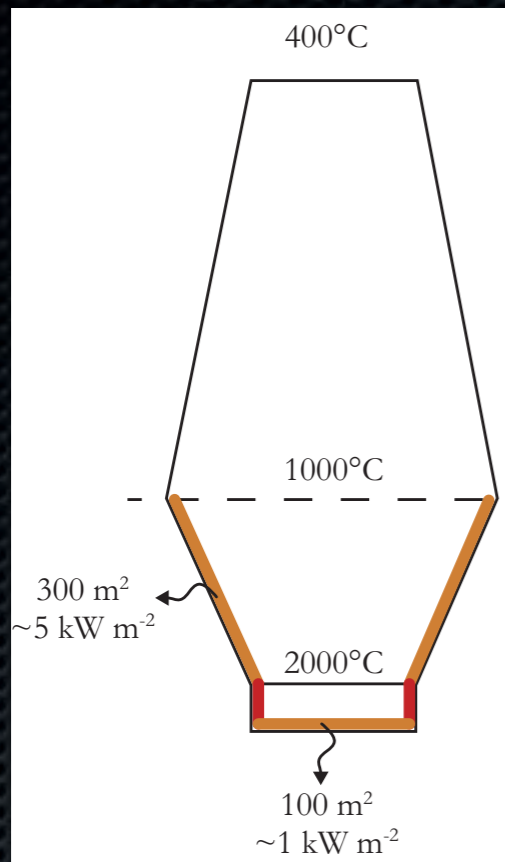
Glass melting, 12m x 20m

image: Sonilas Benelux

# Heat management : glass



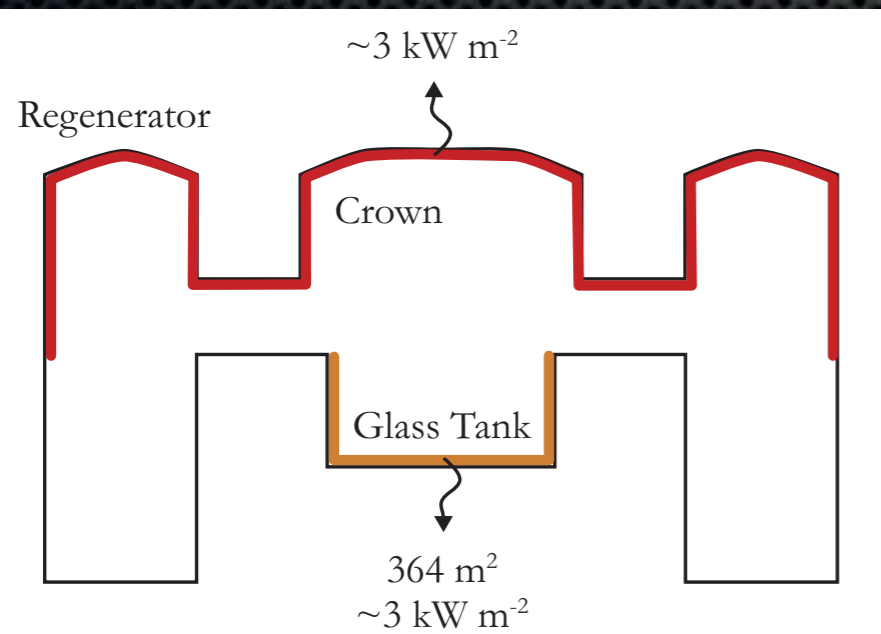
# Heat management : high T !



- Temperature in excess of 1000°C
- Heat flux less than few kW/m<sup>2</sup>
- Surfaces greater than 20,000 sq ft

Is it possible to harvest or manage such heat?

thermoelectric at temperature greater than 1000°C for low heat fluxes?



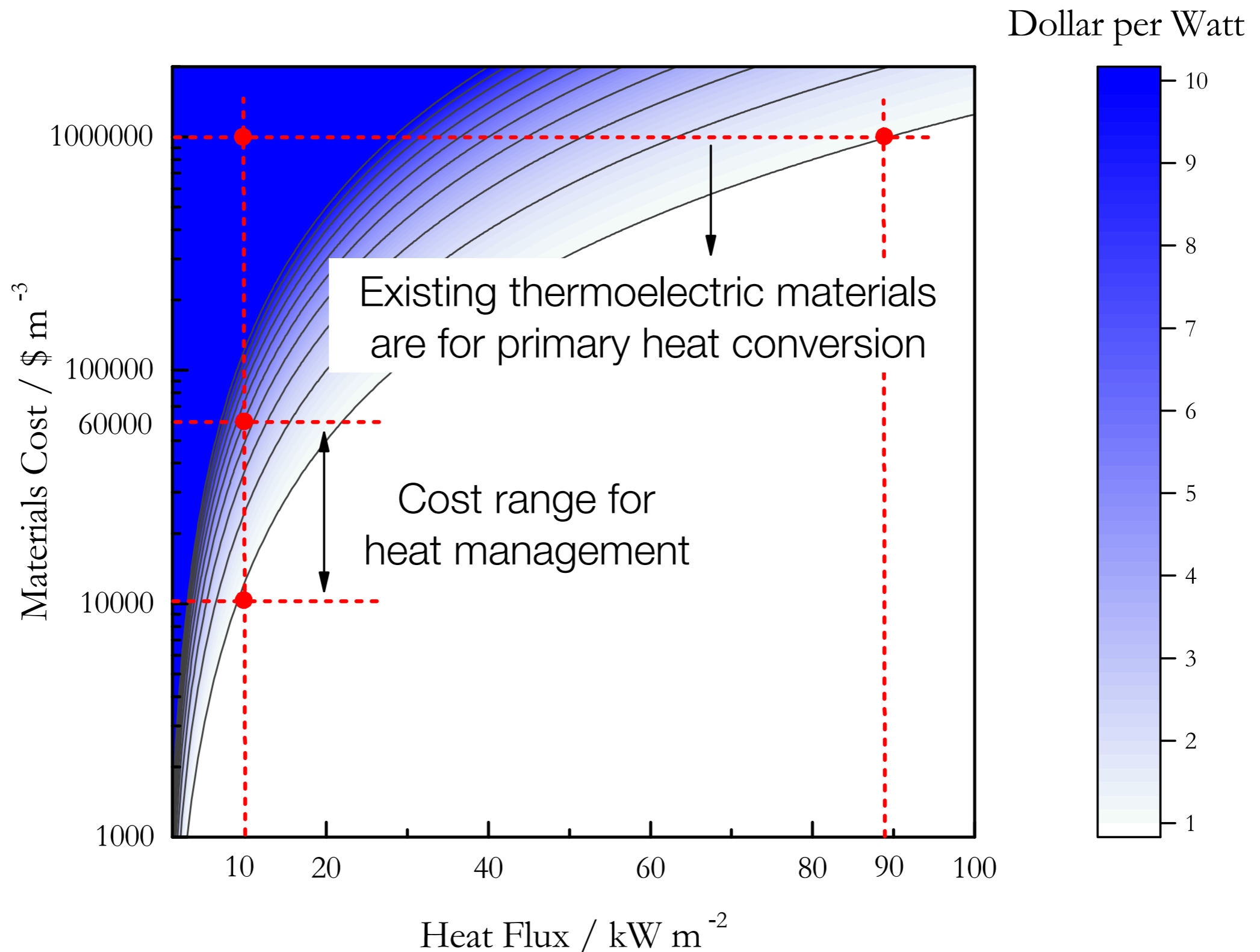
# Thermoelectricity for low heat fluxes

$\Delta T$	thermal conductivity W/m/K	thickness material	heat flux
400	2	1 cm	80 kW/m <sup>2</sup>
400	2	16 cm	5 kW/m <sup>2</sup>

e.g. crystalline Bi<sub>2</sub>Te<sub>3</sub>  
225°C  
few cm thick?

Industrial reality: temperature in excess of 1000°C, few kW/m<sup>2</sup>  
call for different metrics and materials

# Thermoelectricity for low heat fluxes



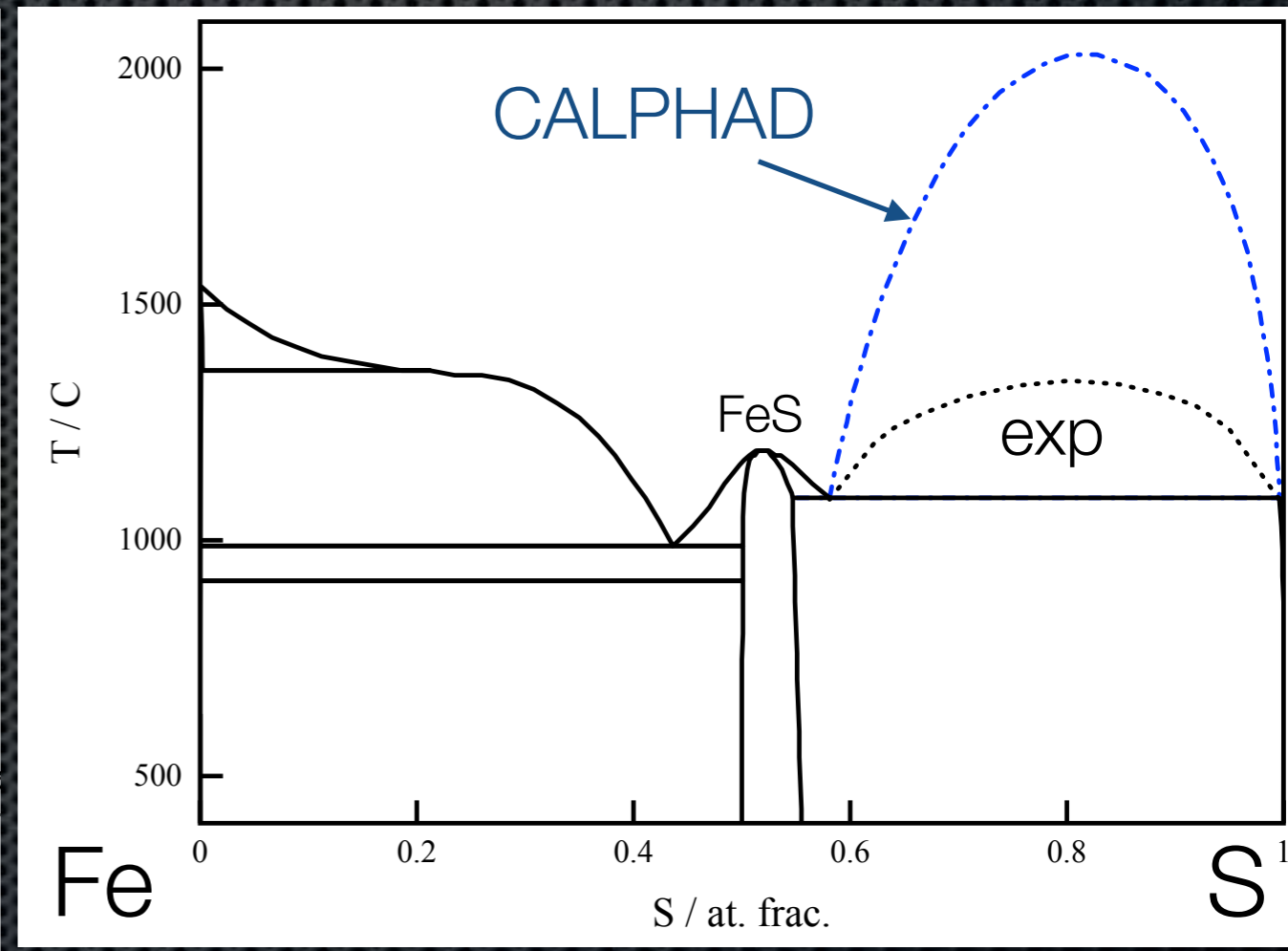
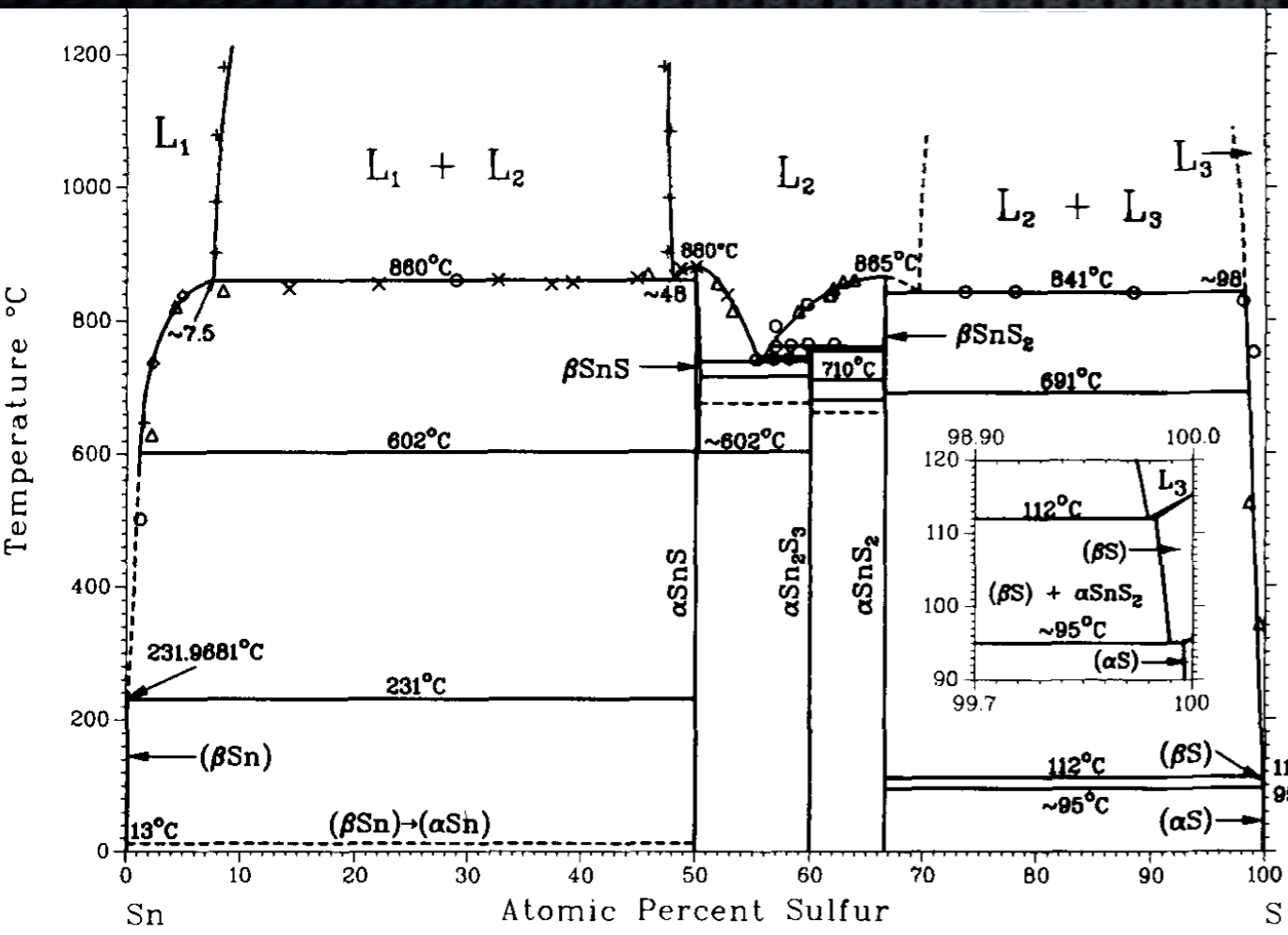
# Molten semiconductor?

	Figure of Merit	Thermal conductivity	Melting point	Cost \$/m <sup>3</sup>
Ni <sub>3</sub> S <sub>2</sub>	na	na	800°C	54,000
SnS	na	na	882°C	65,000
PbS	na	na	1120°C	13,500
Cu <sub>2</sub> S	1 @1130°C	0.8 to 1.4	1130°C	27,400
FeS	na	na	1200°C	<5,000
FeO	na	na	1400°C	<3,000

How do we know the range of temperature for semiconductivity?  
Can we demonstrate power generation at  $T > 1000^\circ\text{C}$ ?  
What are the actual material performance limitations?

# Materials science for high T?

- Handbook of phase diagrams...



Sharma and Chang, Bulletin of Alloy Phase Diagram, 1986



# Materials science for high T?

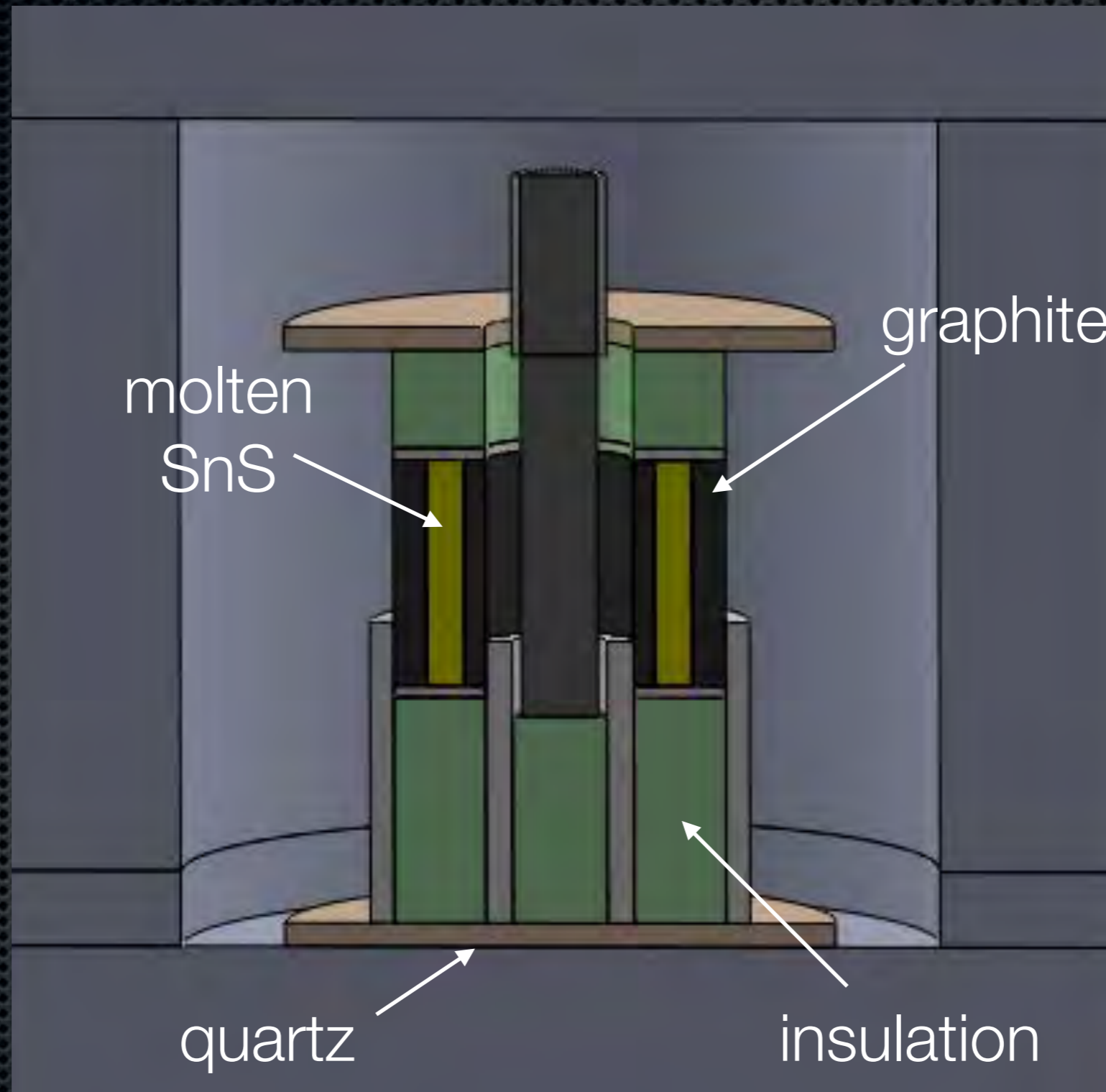
- Need for transport properties: Handbook again?
  - density
  - viscosity
  - electronic properties (conductivity, Seebeck, mobility)
  - thermal conductivity:
    - electronic thermal conduction?
    - radiation? ... optical properties

most useful database are trade secrets...  
and unique for each field (!)

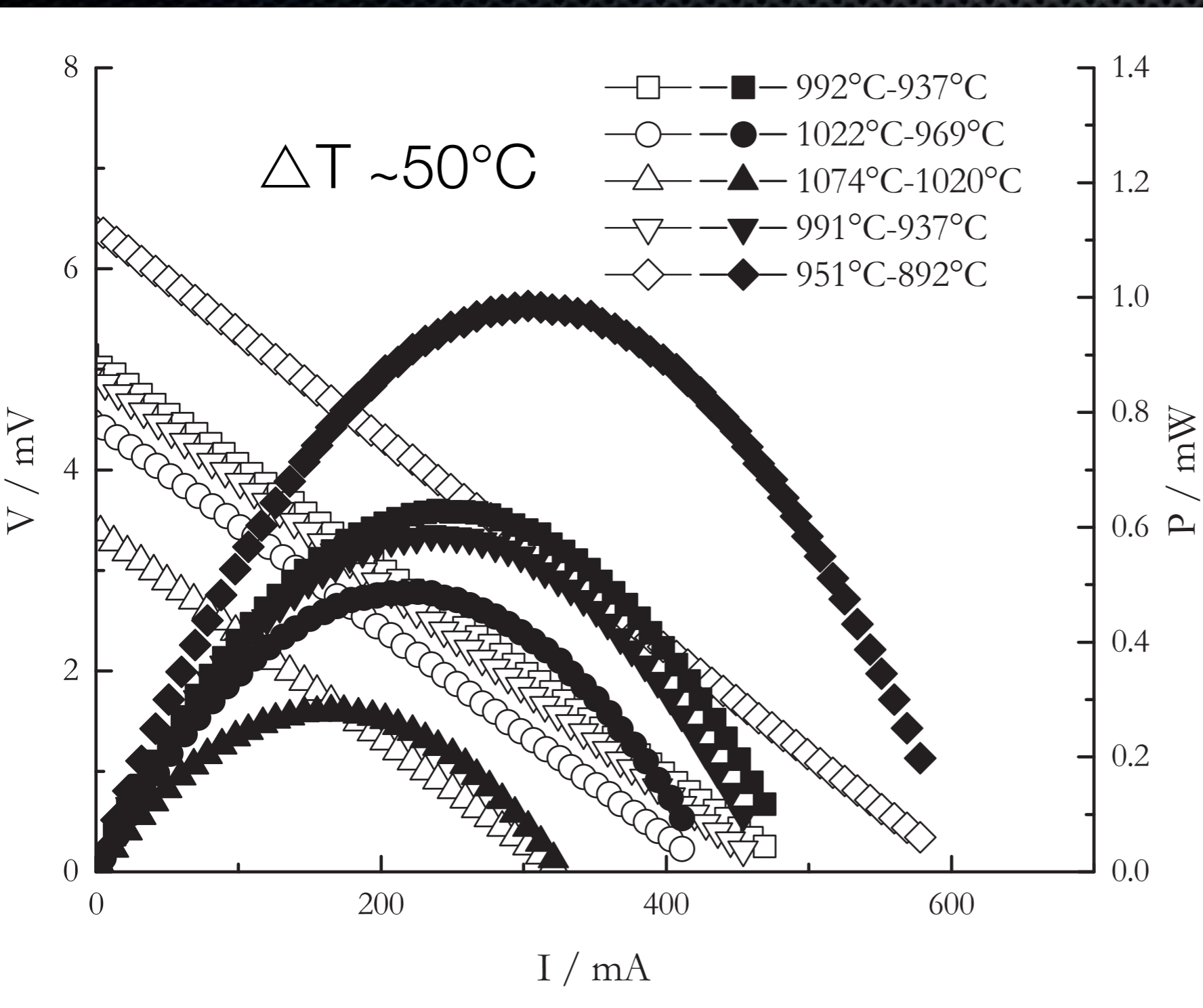
# Acknowledgements



# Device tested with SnS



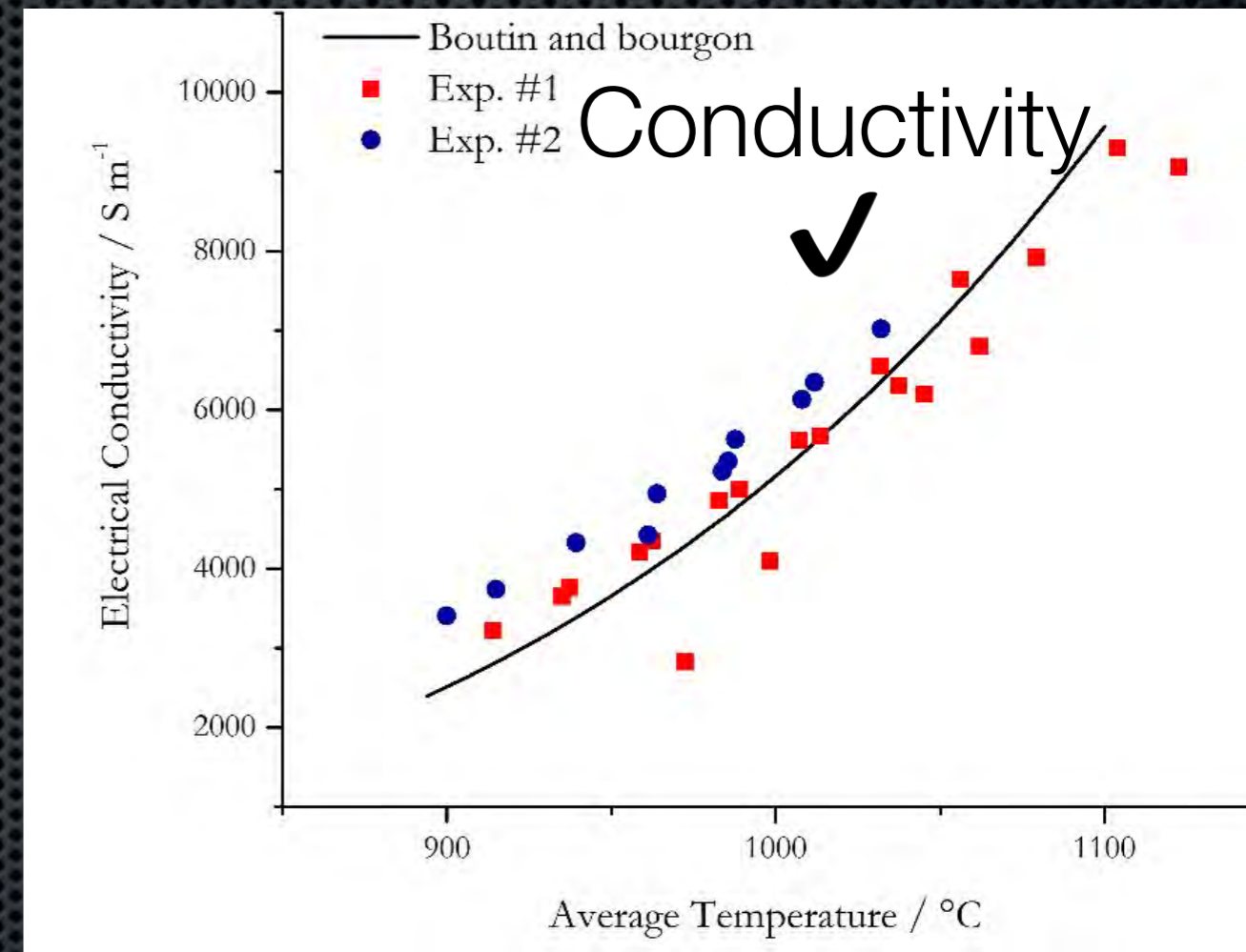
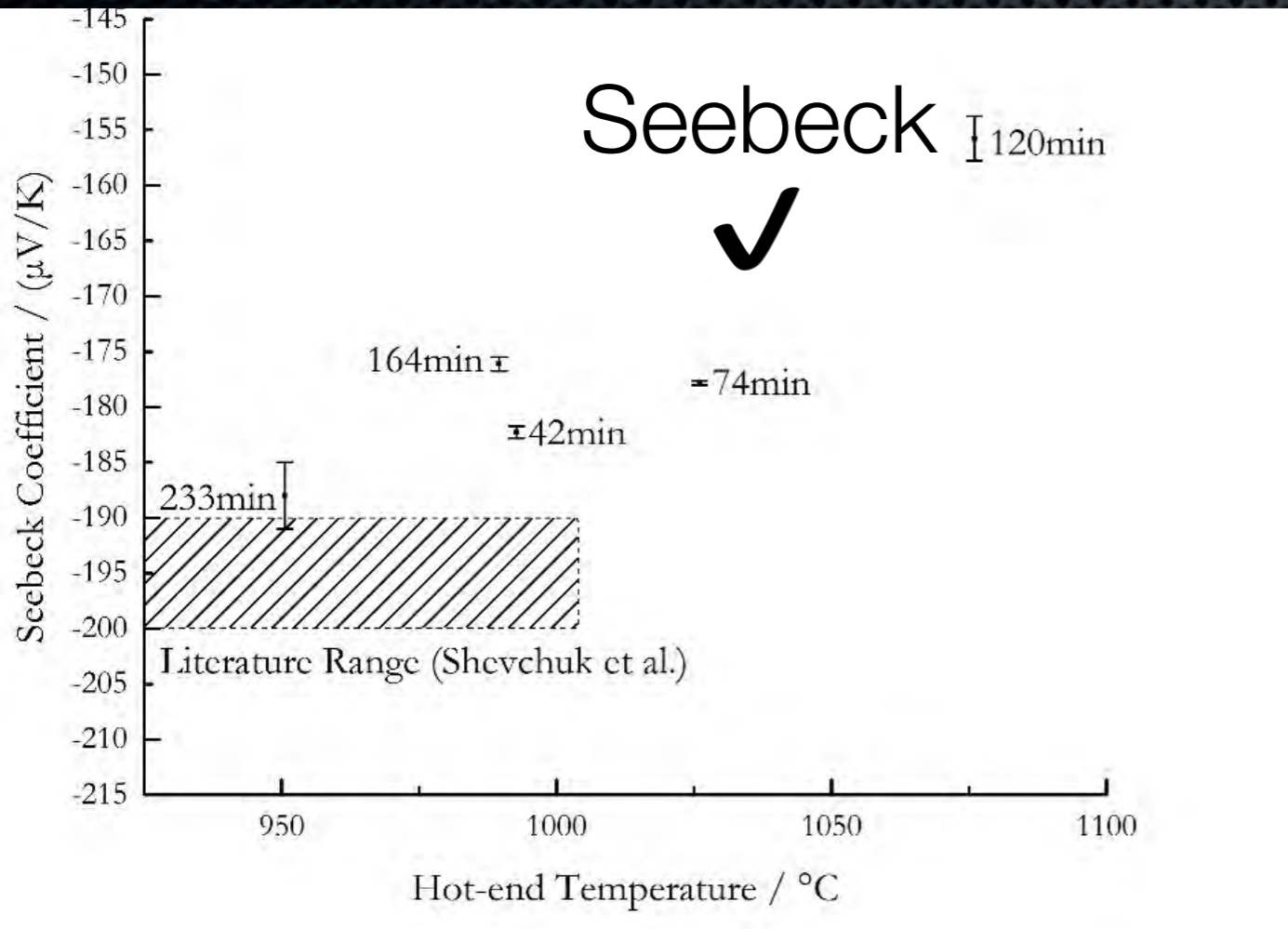
# Device tested with SnS



- successful power generation at  $T > 1000^\circ\text{C}$
- very stable power
- >4h operation
- reusable

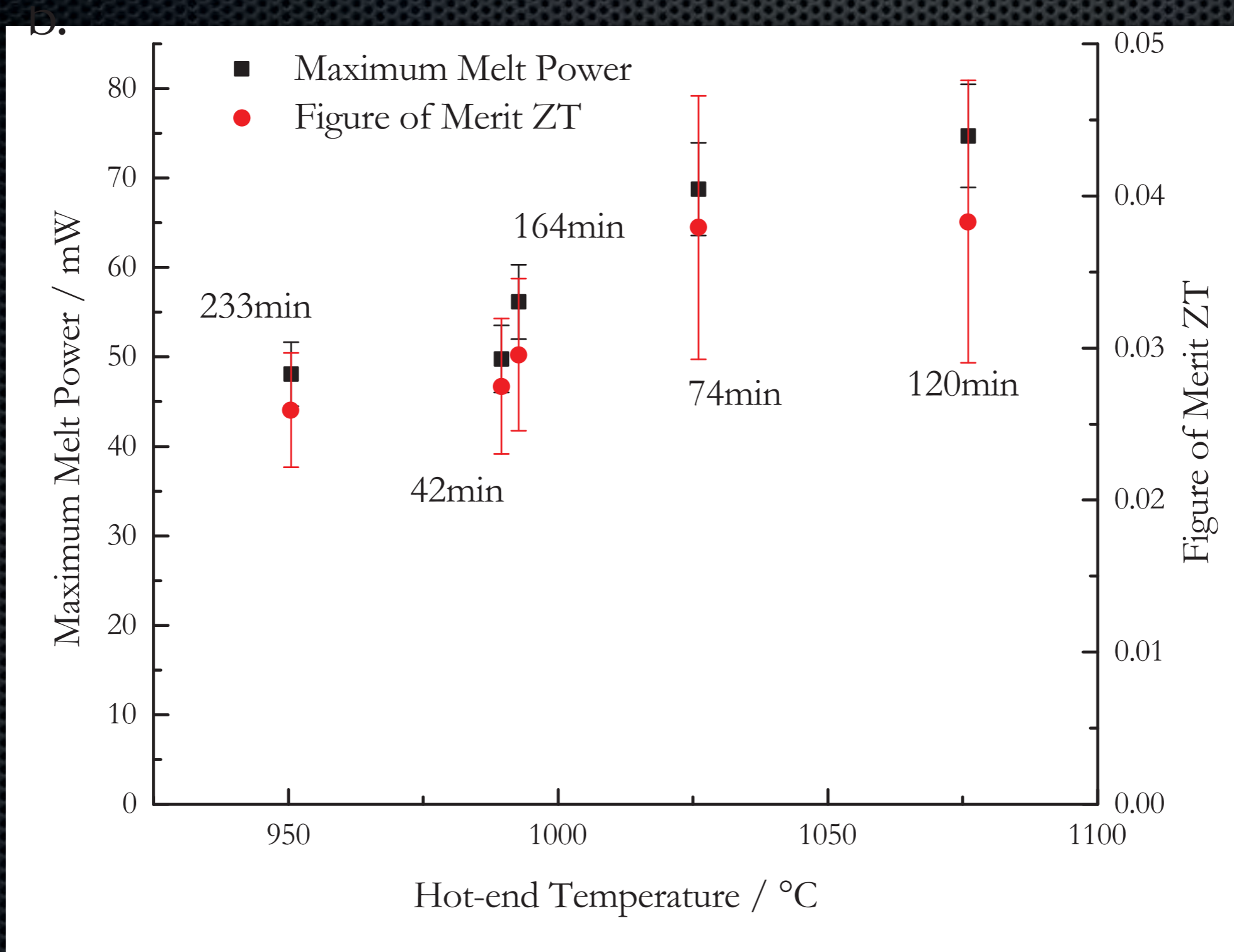
# Device and material

Results in agreement with known electronic properties

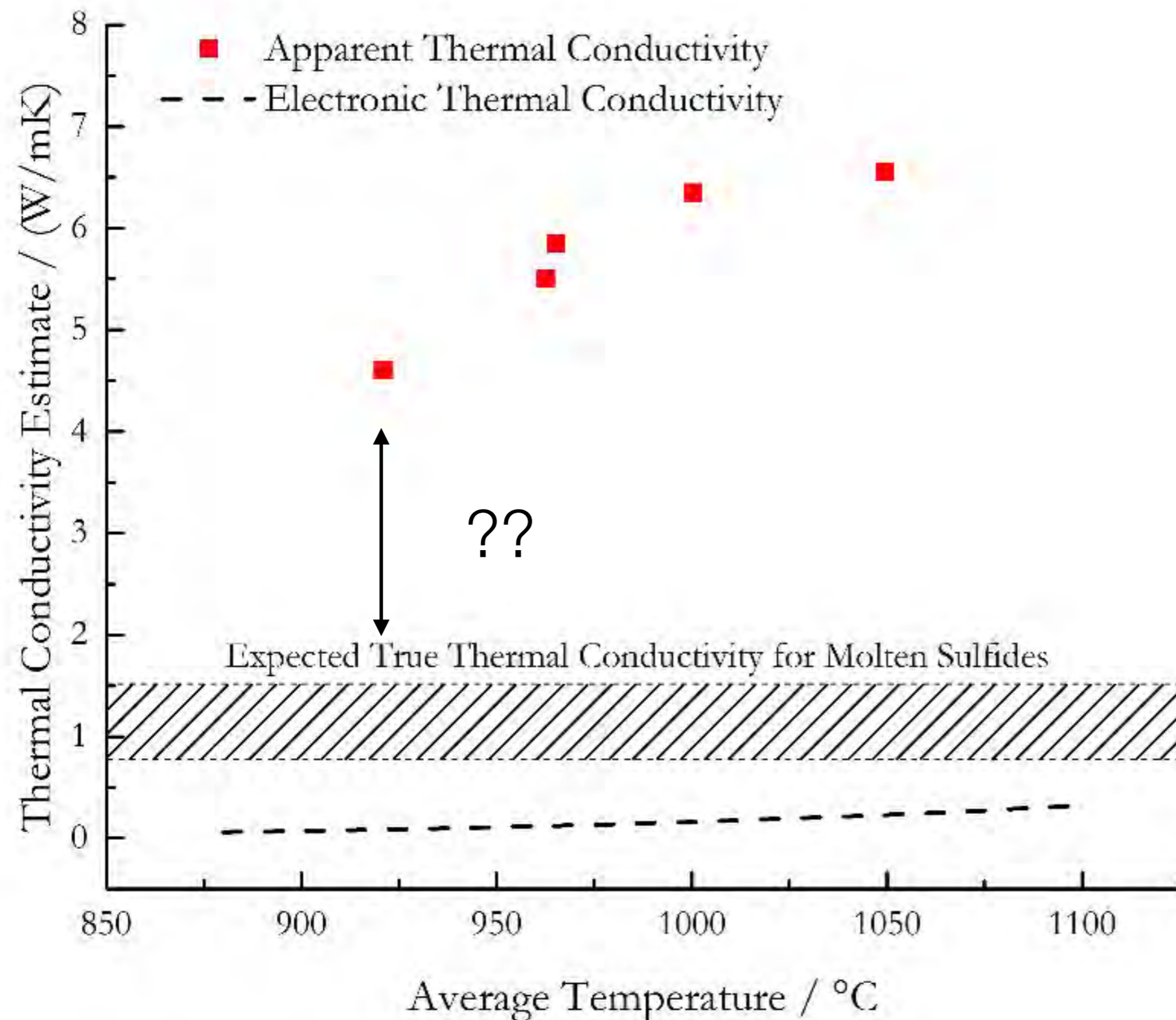


# Materials limitations?

Good cyclability - what limits power conversion?



# Thermal conductivity?



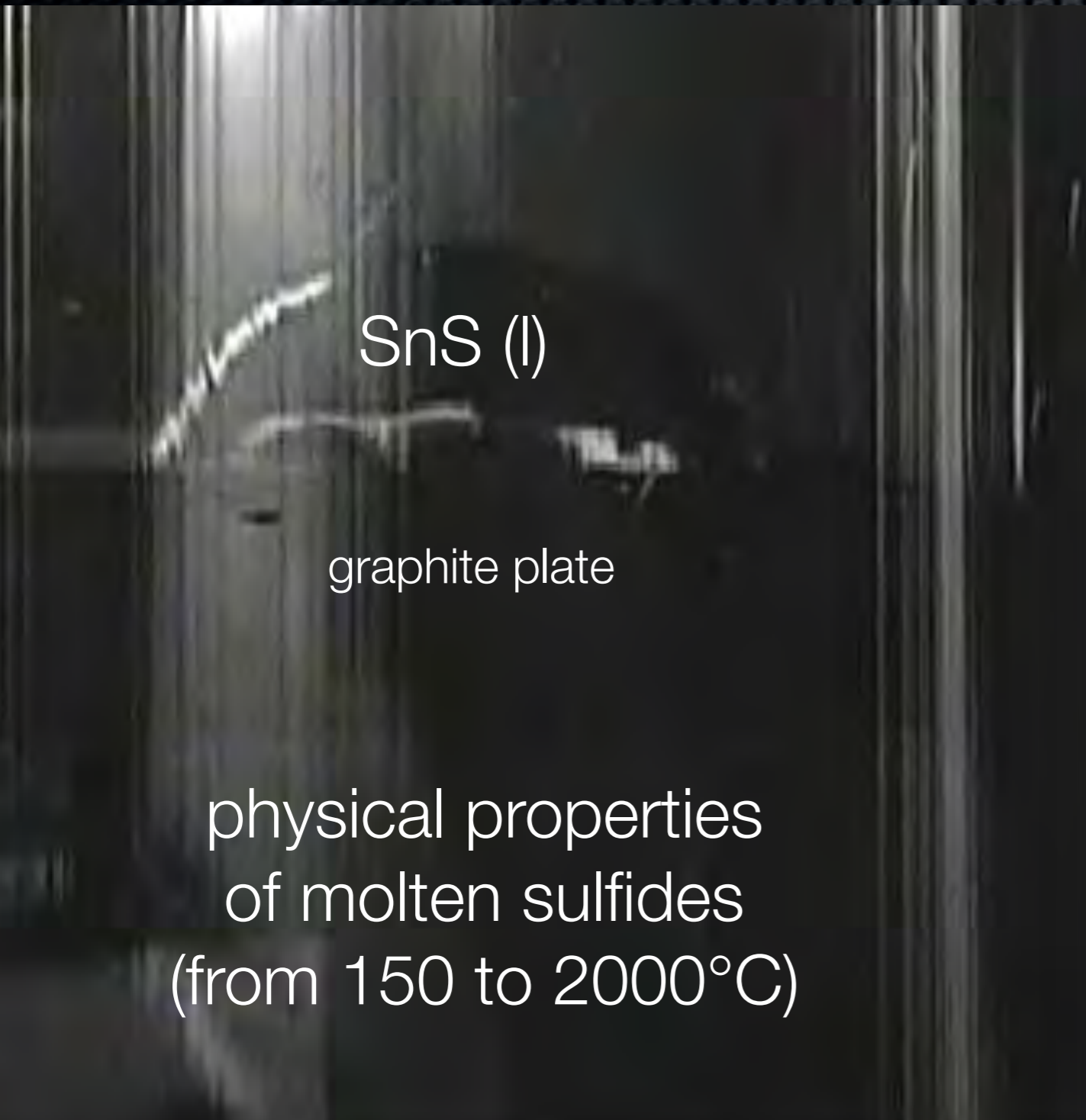
## CONVECTION?

- need density, viscosity data

## RADIATIVE?

- Refractive index  $n$
- Absorption coefficient  $\alpha$

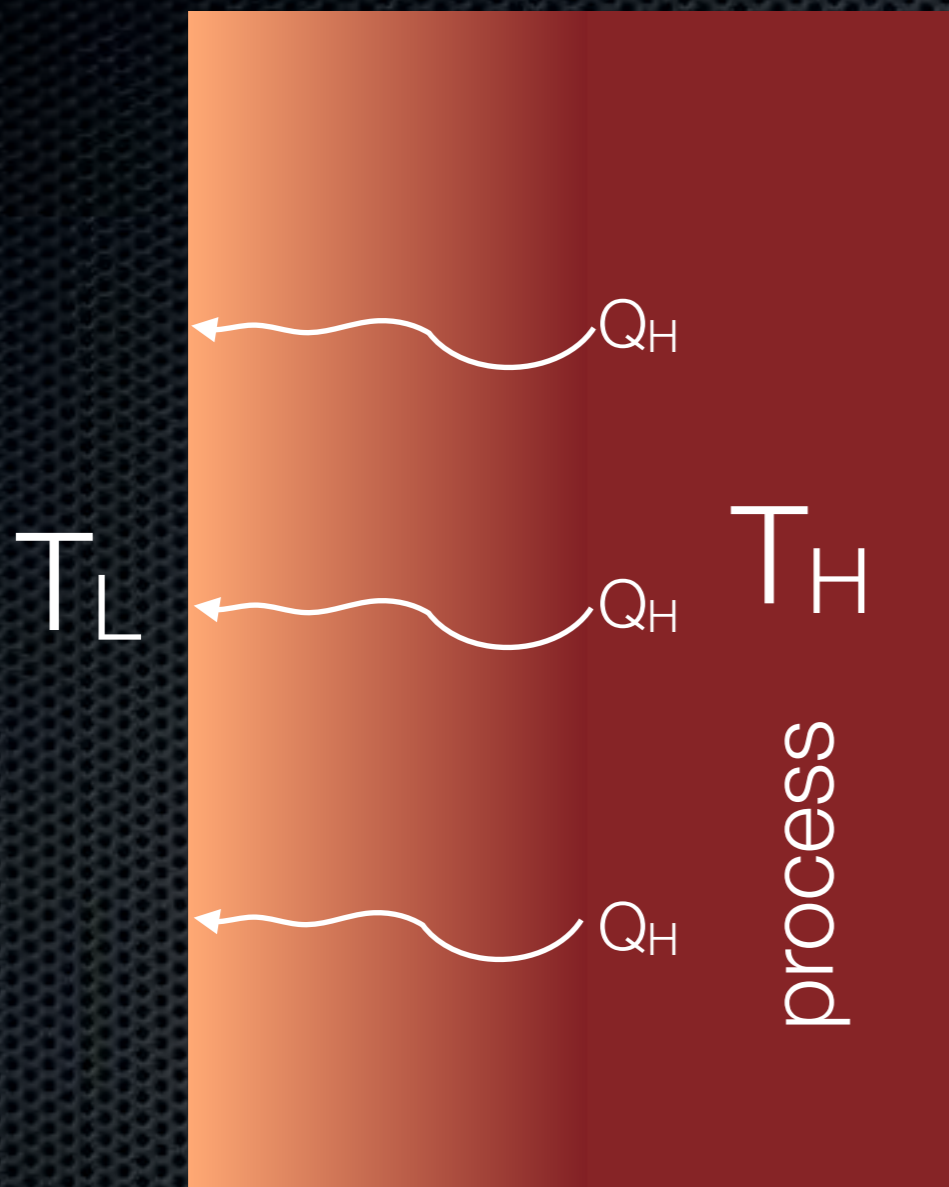
# Seeing the melt



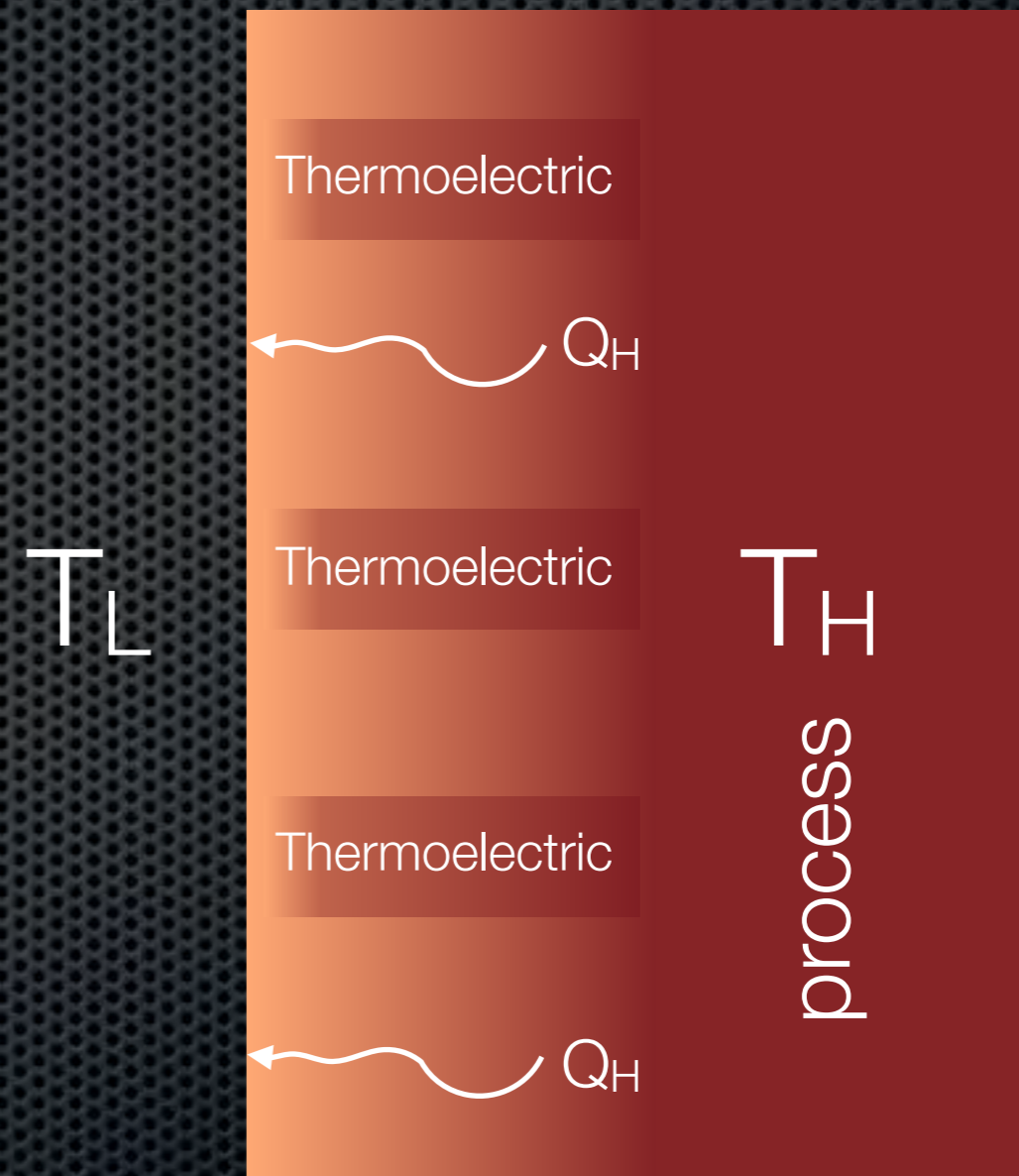


# Process heat management?

conventional refractory



active refractory



- $Q_H$  is fixed by max  $\Delta T$  of refractory
- $T_H$  limited by  $T_L$  location and ability to remove  $Q_C$

- higher apparent thermal conductivity
- ability to be controlled by electricity
- can enable higher  $T_H$ , fast  $Q_H$  removal

# Summary

- High temperature melts represent the ultimate state of condensed matter, pertinent to the entire metal supply chain (including oxidation products!)
- Existing predictive and experimental framework for properties is quickly evolving
- Novel media for heat, electrical or mass transfer in high temperature conditions can be designed and scaled-up promptly
- Scalability and implementation will be possible if it increases productivity of underlying process flow-diagram



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# Acknowledgements

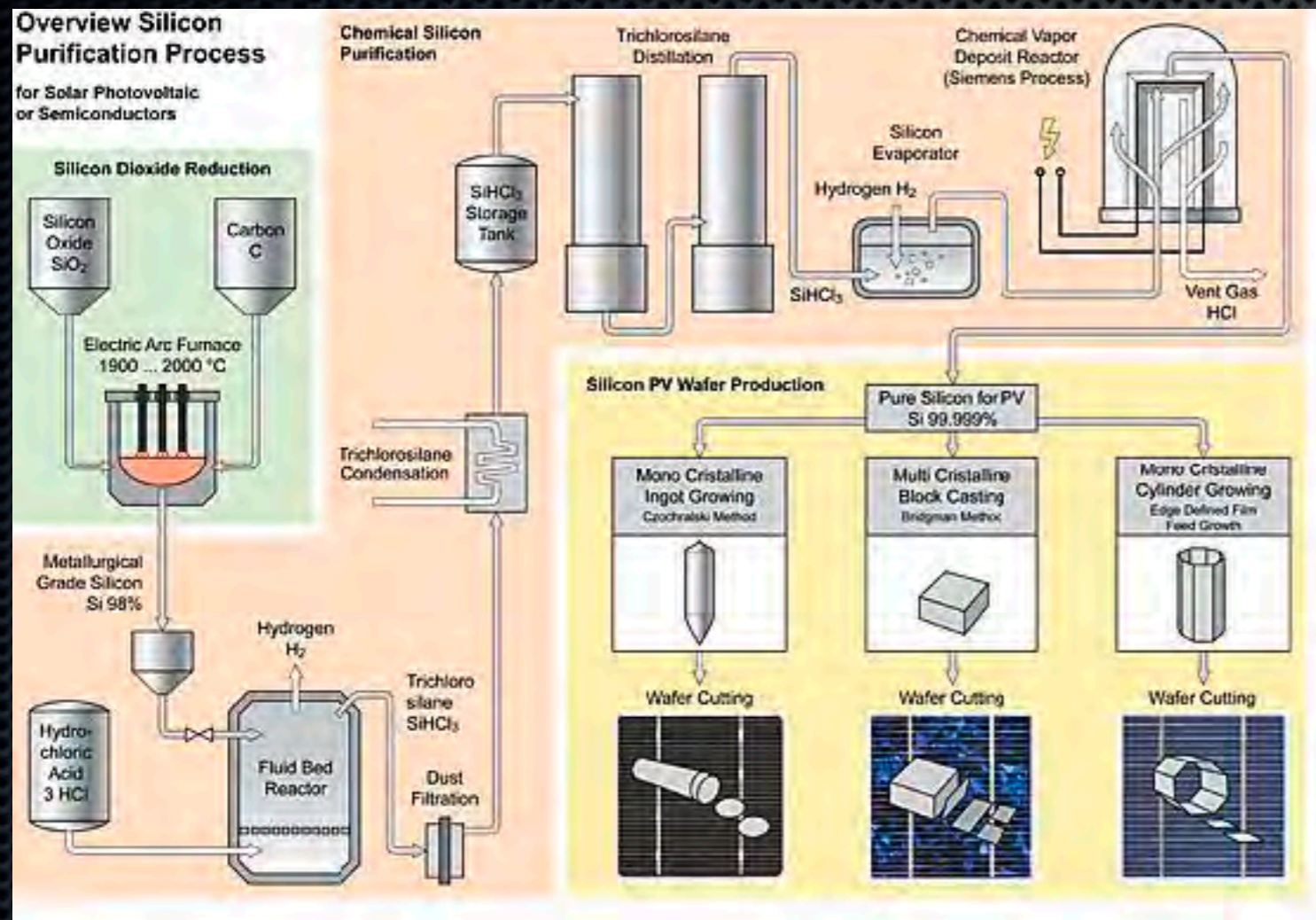


Dr. Cooper Rinzler - Mr. Youyang Zhang



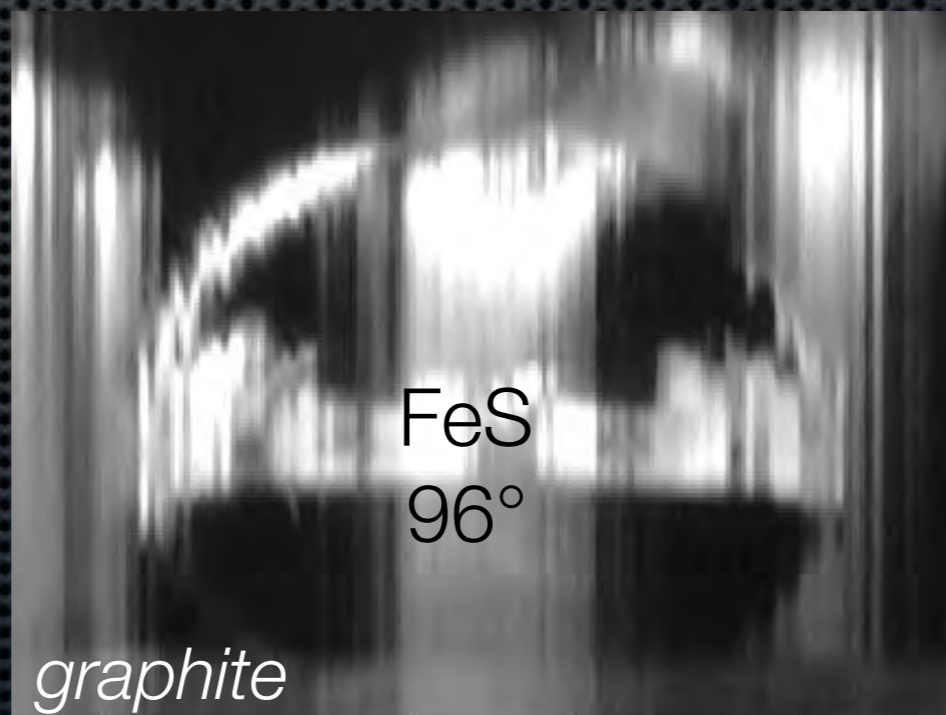
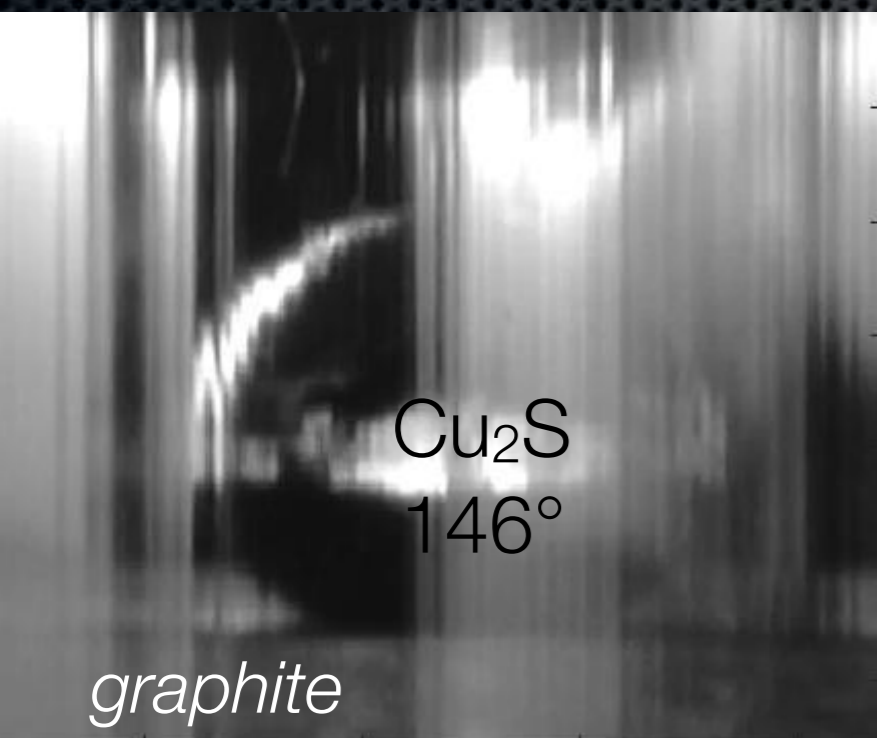
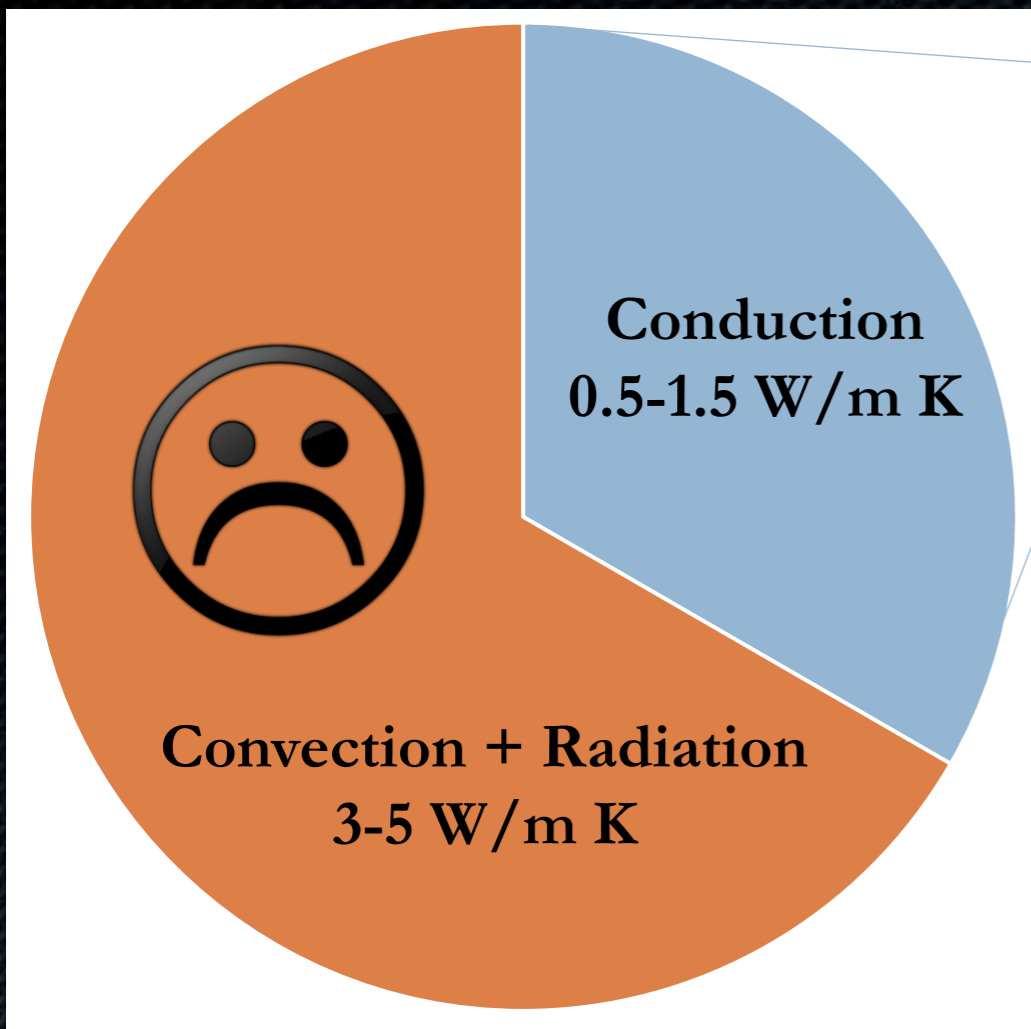
# Materials processing

- realm of liquid state, high temperature, plasma, ...



Electric Arc Furnace,  
Ingot,  
Plasma deposition,  
EBeam (sublimation),  
RT Process...

What is the state of the art in 'high temperature' materials science?



# Thermal conductivity?

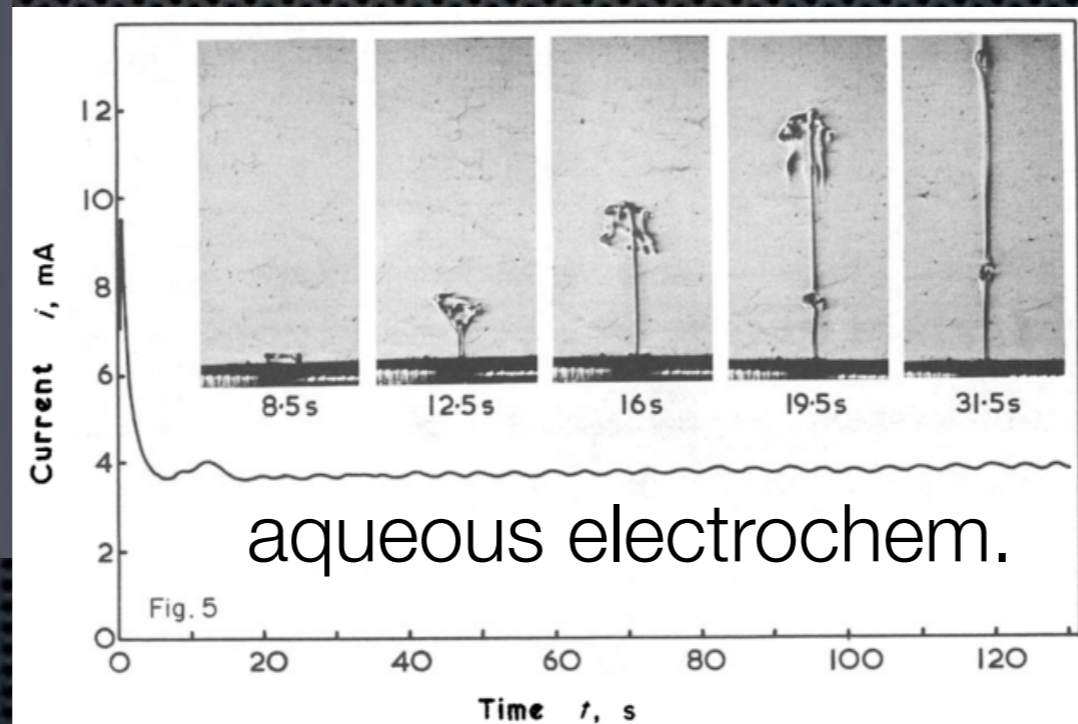
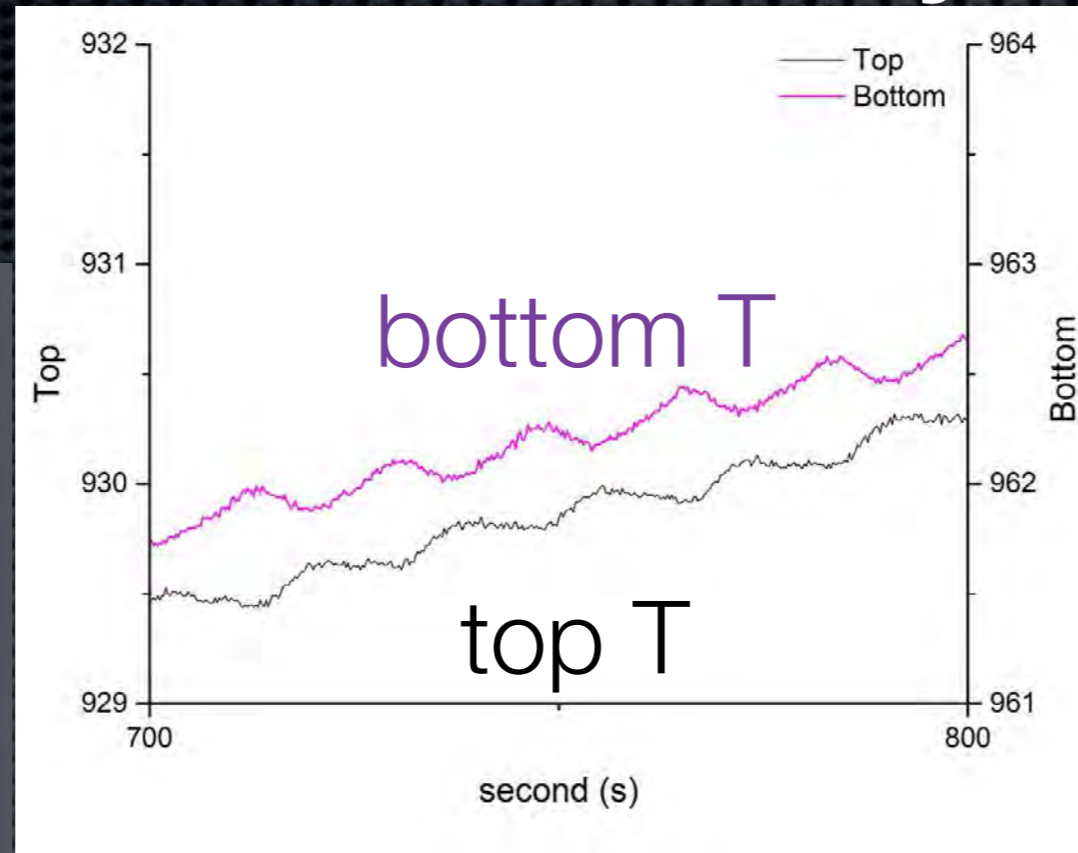
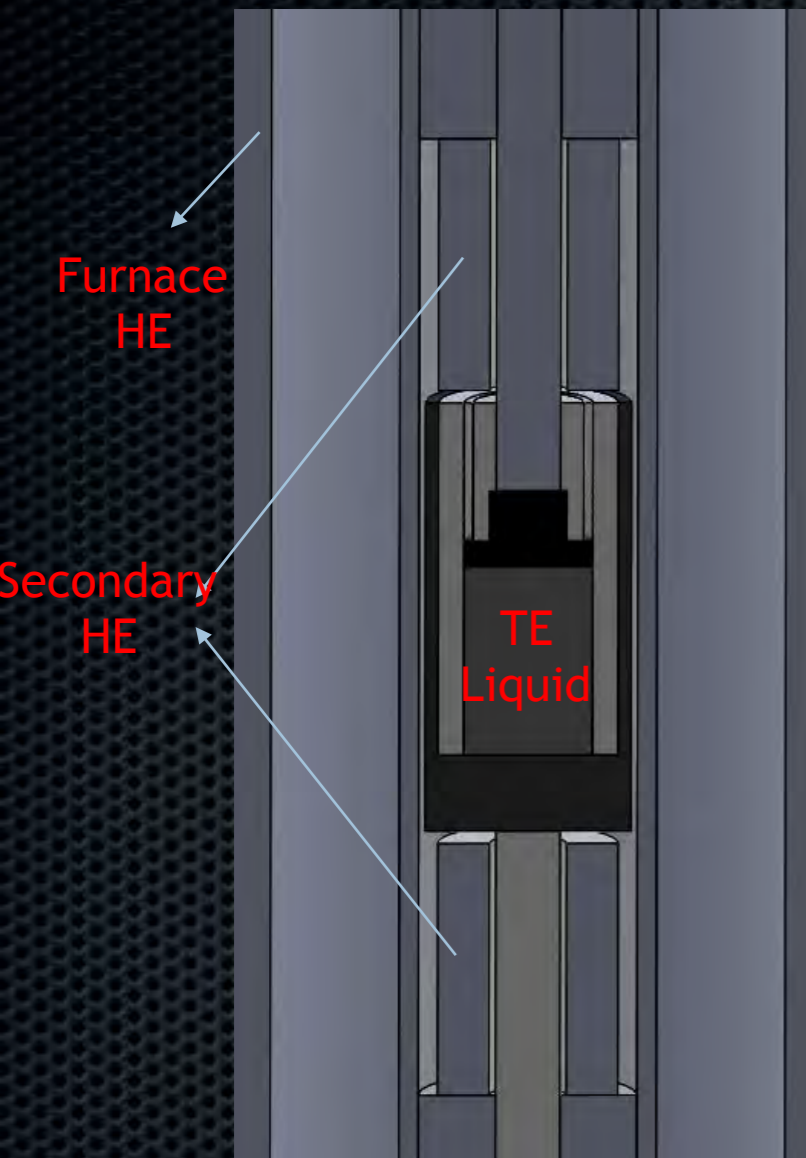


FIG. 5. Current-time plot and convection development for 2.5 mm electrode in 0.3 M solution,  $Ra_d = 1.0 \times 10^7$ .

M. A. Patrick and A. A. Wragg, *Int. J. Heat Mass Transf.*, **18**, 1397-1407 (1975).

natural convection?



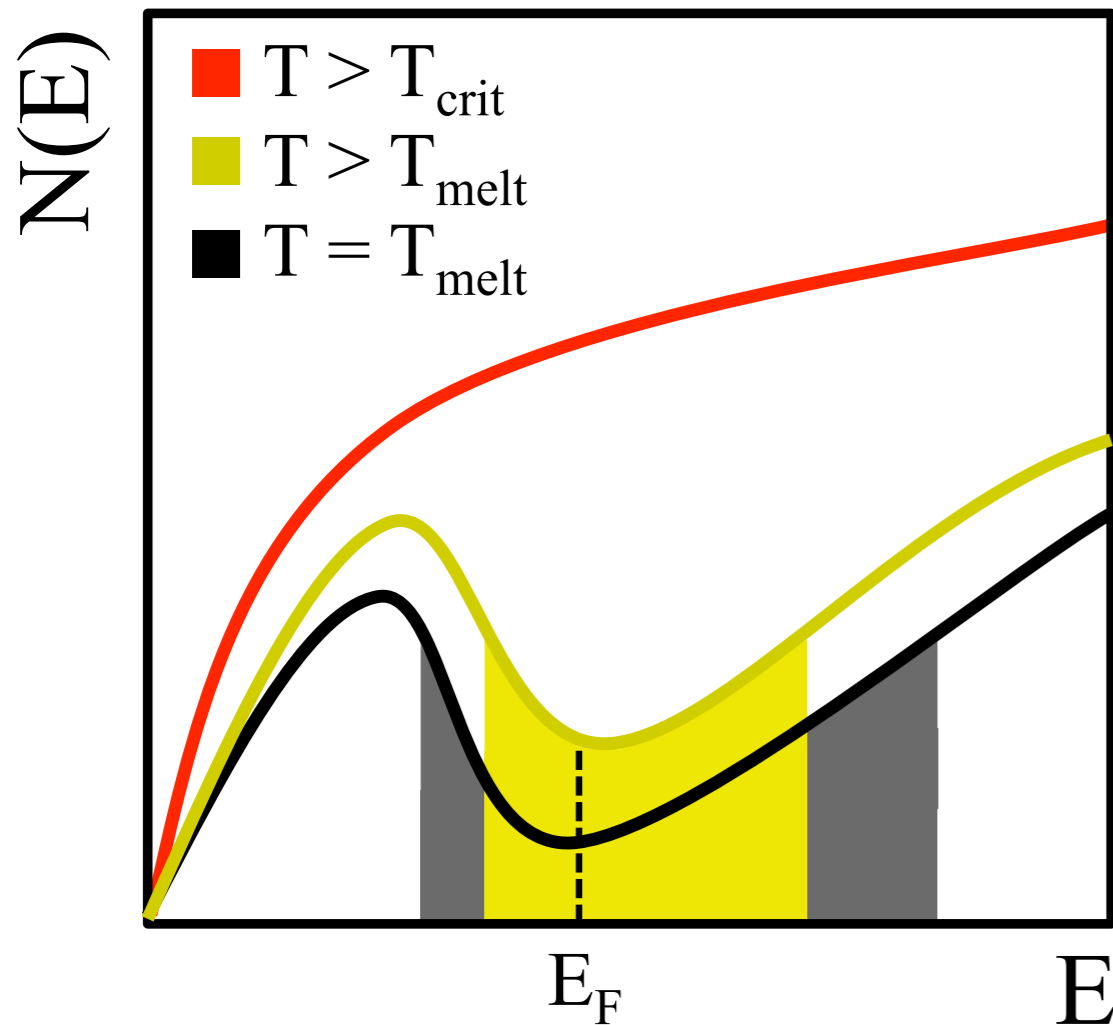
insight into density, viscosity, diffusivity



seeing the melt?

# Liquid semiconductors

Certain compounds maintain semi-conductivity upon melting...described by a 'pseudo-gap'



Mott's formulas  
for amorphous SC

conductivity  $\sigma = \sigma_{min} e^{\frac{-(E_0 - \gamma T)}{kT}}$

thermopower  $\alpha = -\frac{k}{e} \left( \frac{E_0}{kT} - \frac{\gamma}{k} + 1 \right)$

J. E. Enderby and A. C. Barnes (1990), Rep. Prog. Phys., 53, pp. 85–179

V. A. Alekseev, A. A. Andreev, and M. V Sadovskil, (1980), Sov. Phys. Usp., 23, no. 9, pp. 551–575

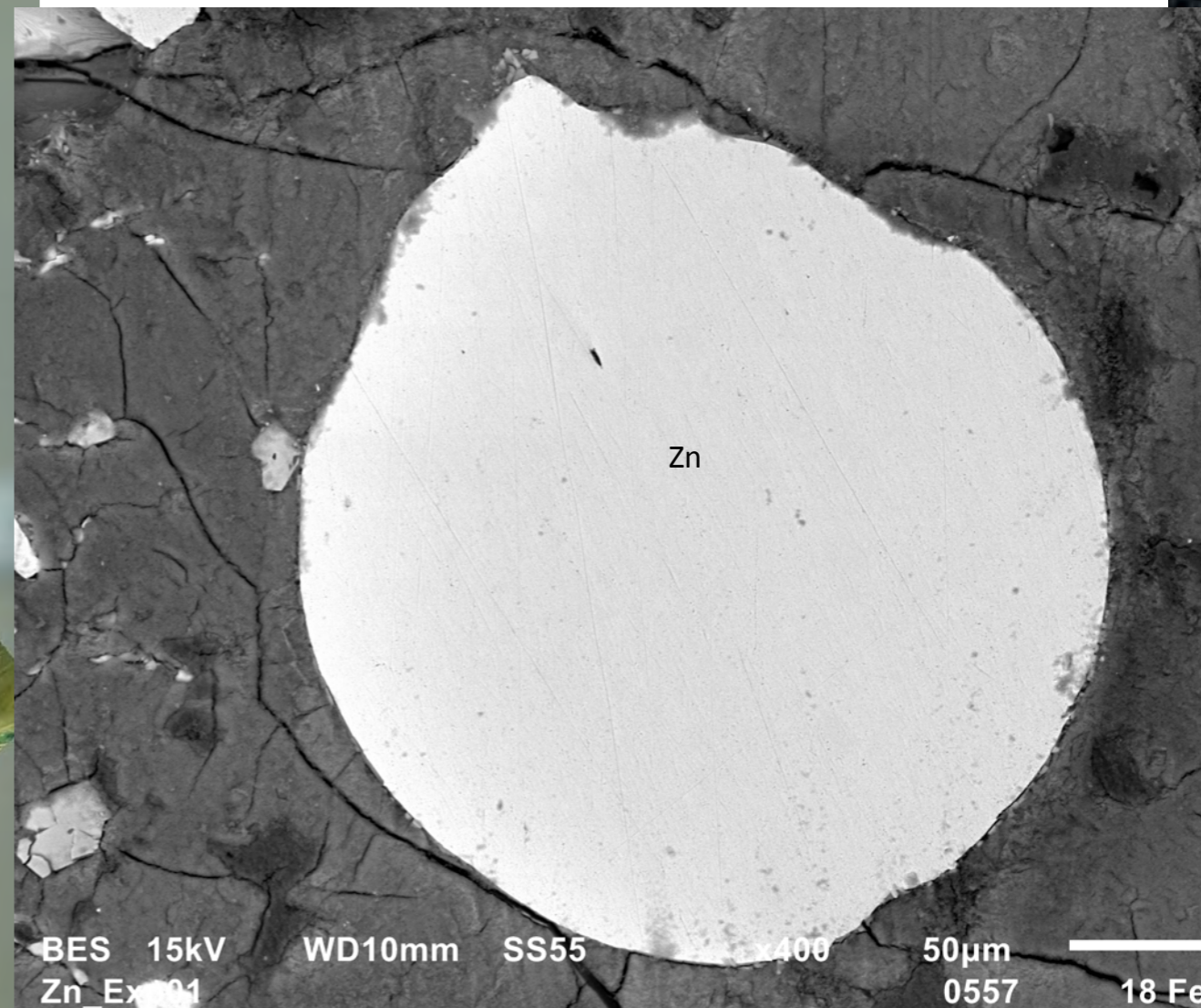
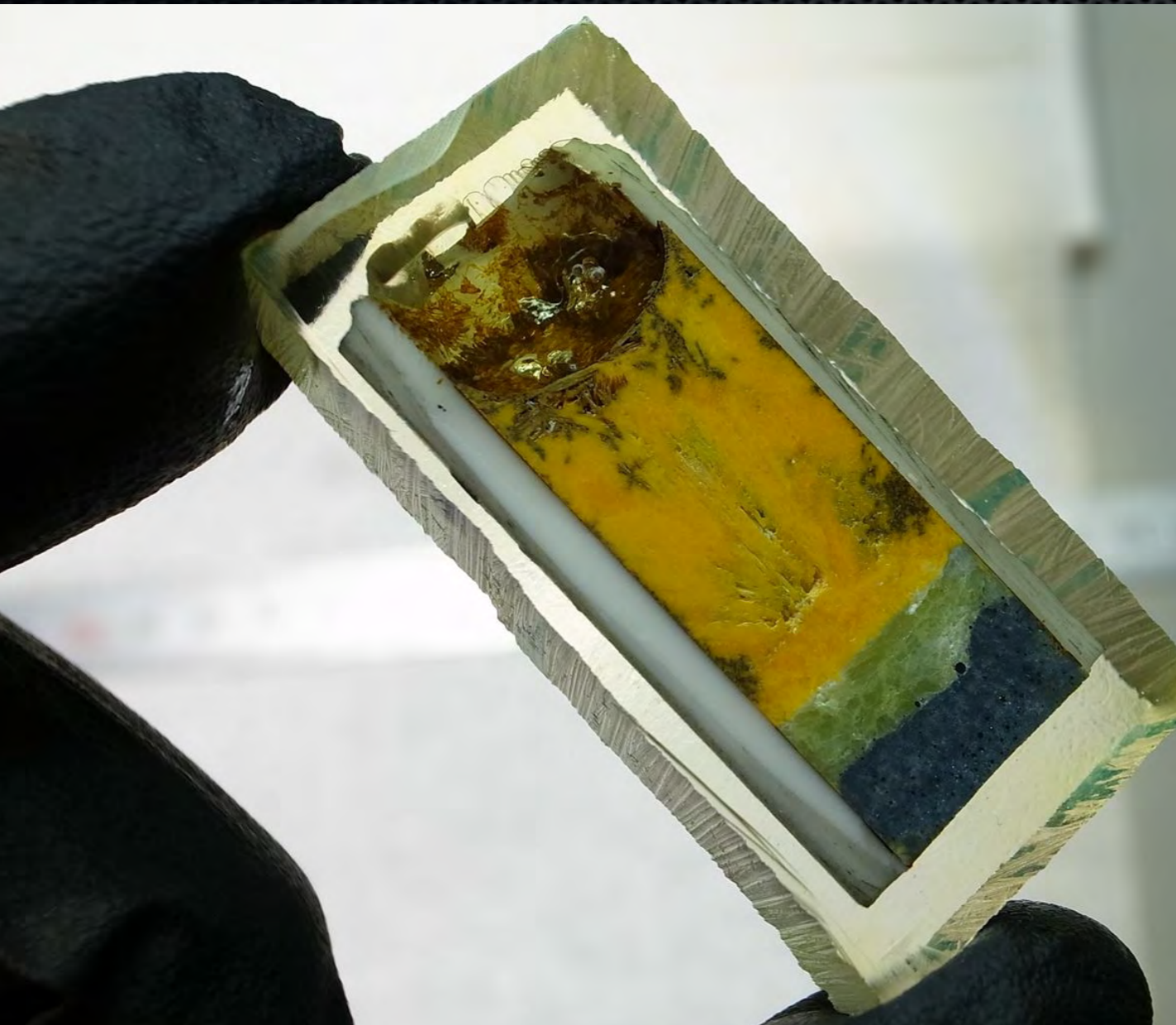
D. L. Price, (1990), Annu. Rev. Phys. Cher

P. W. Anderson, (1958), Phys. Rev. 109, pp.1492– 1505,



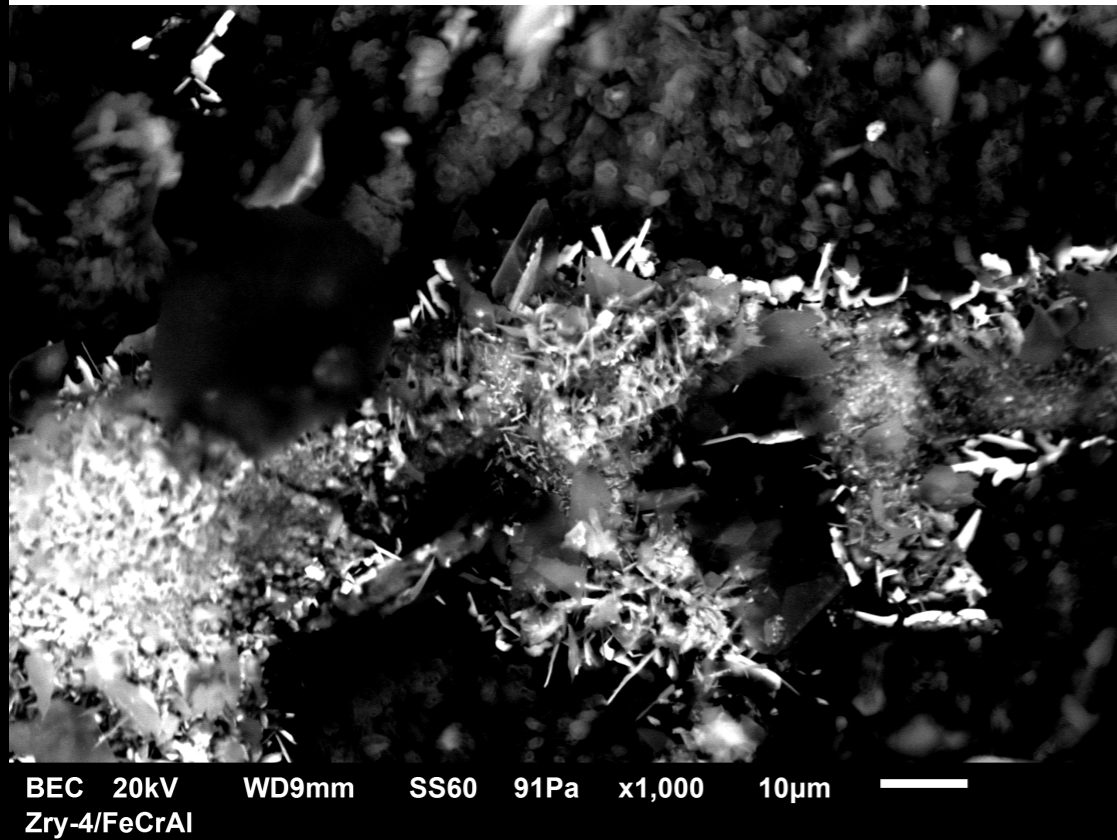
# Molten sulfide electrolyte #2

$\text{Na}_2\text{S-ZnS}$



Allanore et al., Presentation at the 11th Workshop on Reactive Metal Processing  
February 20, 2016, Cambridge, MIT

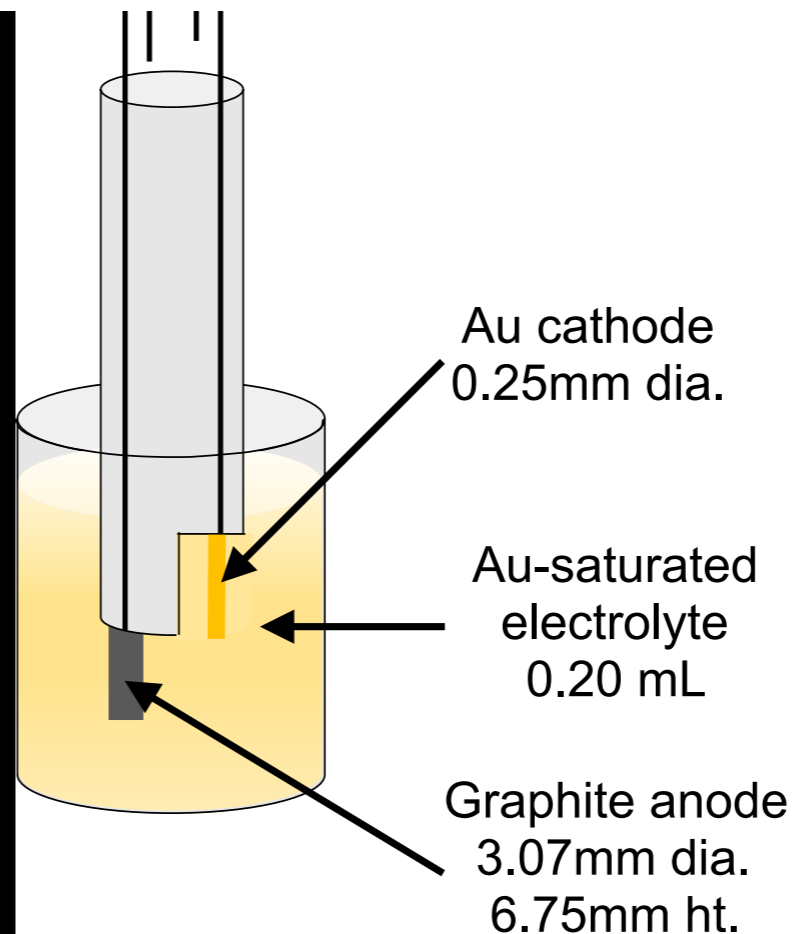
# Silver: estimated concentration of 135 g/L



After 115 minutes of electrolysis, Ag "dendrites" were found near the cathode

additional possibilities demonstrated with  $\text{Na}_2\text{S-ZnS}$

# Gold: estimated concentration 278 g/L



Wagner & Allanore, Poster at the 68th International Society Meeting August 28, 2017, Providence (RI)